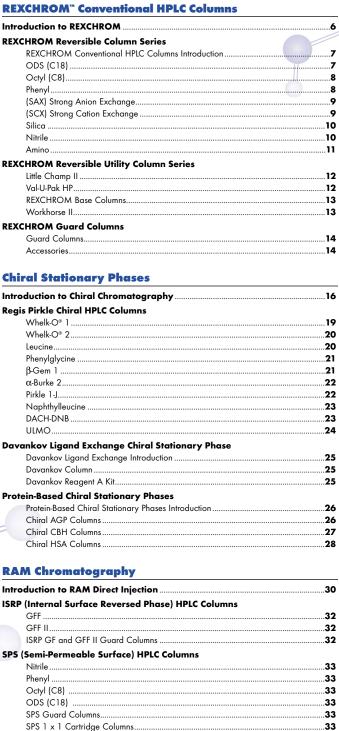
CHROMATOGRAPHY CATALOG

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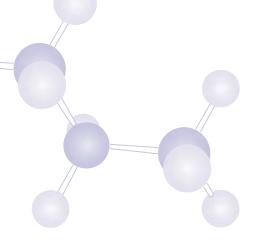




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REGIS TECHNOLOGIES, INC.

Chromatography

Regis has a long tradition of serving the analytical needs of scientists and researchers worldwide with its vast array of chromatography products and technical assistance. Regis manufactures an extensive line of chromatography stationary phases and high purity

GC derivatization reagents. Regis is the exclusive manufacturer for the Pirkle-Type Chiral Stationary Phases for analytical, semi-preparative and preparative applications and separations.

Regis is also the exclusive manufacturer of many other specialty HPLC phases.

Research and development are essential parts of our chromatography business that allows us to introduce new and innovative products that meet the needs of our customers.



proud of its continued growth
and leadership role in its three
areas of business:

- technically advanced chromatography products
- custom bulk
 pharmaceuticals and
 other fine organic
 chemicals
- chromatographic separations.

Custom Synthesis

In addition to chromatography products, Regis manufactures Active Pharmaceutical Ingredients (APIs) and intermediates on a custom basis. Regis is committed to meeting the customer's needs through initial process development, scale-up and ongoing production. Our

facility is run according to current Good Manufacturing Practices (cGMP) and is routinely inspected by the FDA. All pharmaceutical intermediates and final products are manufactured and tested to customer specifications. All technology developed for the customer is owned by the customer and all batch records are provided to the customer as well. In addition to APIs, Regis also produces and catalogs a number of fine organic chemicals. For a complete list, check our Web site at www.registech.com/portfolio.html.



Regis' production department includes a pilot plant dedicated to cGMP chromatography. Glass gravity columns are available in diameters of 4", 6", 9" and 12", with capacities of up to 50 kg of packing material. In addition, Regis has some preparative HPLC capability. We are currently limiting the use of the chromatography equipment to projects involving both custom synthesis and separations. Regis no longer accepts projects requiring only separations.

REGIS TECHNOLOGIES, INC.

Expanded Manufacturing Facility

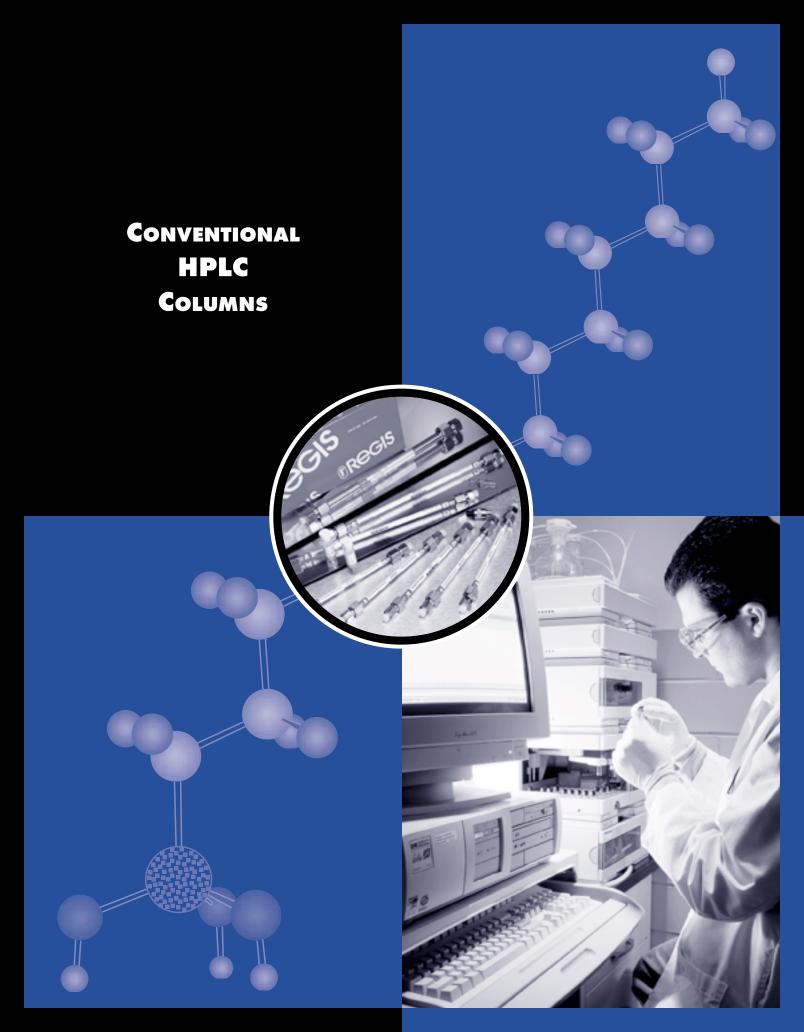
The management at Regis continuously reinvests in the company. Between 1995 and 2001, Regis expanded and improved every lab and manufacturing area. This included a new production facility with six dedicated reactor suites, individual kilo lab suites, an expanded quality control department and the installation of a cryogenic reactor. Plans are underway for increasing capacity with additional production labs, individual kilo lab suites and reactor suites. The net result is repeat business from satisfied customers and low turnover because of satisfied employees. For updates on our expansion, check our Web site at www.registech.com/gmp/.

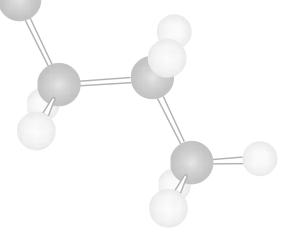
Quality Products and Services

Our high purity reagents and HPLC columns are manufactured on-site according to controlled manufacturing procedures and must meet strict quality control specifications before release. A full

customer service and sales staff is available to answer questions and take orders. Regis also has a complete applications laboratory and knowledgeable technical support staff. With years of chromatography experience, our support staff is dedicated to assisting customers with method development, and column or reagent selection.

Regis is committed to its customer. With this in mind, you can expect Regis Technologies to provide the highest quality products, services and technical support as it continues to grow and meet the challenges of the future.





In an effort to bring ruggedness and reproducibility to liquid chromatography, Regis pioneered the first reversible flow HPLC column. The advent of reversibility brought forth a more uniform, efficient, highly reproducible, and dependable column with extended life. Today, Regis continues to manufacture reversible flow columns in a wide variety of bonded phases.

CONVENTIONAL HPLC COLUMNS

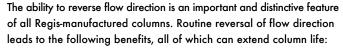


Regis applies its wealth of synthetic expertise in developing bonded stationary phases with superior chromatographic properties.

The REXCHROM stationary phases, with all parameters independently controlled and highly reproducible, satisfy our customers' demands for efficiency. Both 3 µm and 5 µm diameter particles have 200 m²/g surface area, 100Å mean pore diameters and 0.5 cc/g pore volume.

The First Reversible Flow Column

Regis introduced the first truly reversible HPLC column in the early 1980s. Until that time, HPLC columns routinely carried specific flow direction indicators, and users were cautioned that flow in the opposite direction would destroy the column.



- Cleaner outside frit surfaces—reverse flow washes away dirt from the entrance frit
- Deposit-free columns—reverse flow, especially with pure solvents, dissolves and removes column deposits that may have built up near the column entrances
- Fine-free inside frit surfaces—periodic flow reversal holds fines near their point of origin rather than allowing them to accumulate at the exit frit.

All Regis columns are manufactured using an exclusive packing process resulting in an extensive line of high quality, highly reproducible HPLC columns available in a variety of column lengths, particle sizes, and bonded phases.

Quality Assurance

Regis manufactures its columns to the highest quality standards. Each column is packed and tested in accordance with specific control standards. Column performance reports along with care-and-use guides are included with each column.

For questions regarding applications, method development, or custom columns, contact Regis' Sales or Technical Service Department, or e-mail us at sales@registech.com.



REXCHROM Conventional Columns

From its introduction as the first explicitly reversible HPLC column, the Regis REXCHROM Reversible has been the column of choice for efficiency, dependability, quality, and extended life. Each REXCHROM phase has a very uniform bonding chemistry. The specifications for the REXCHROM Column Series are detailed in the adjacent chart. These columns have demonstrated effective separations in a variety of applications.

Specifications Phase Loading, μmol/m²			
Phase	5 μ m	3 μm	Endcapping
ODS (C18)	2.9 ± 0.3	2.9 ± 0.3	Trimethylsilyl
Octyl (C8)	3.2 ± 0.3	3.2 ± 0.3	Trimethylsilyl
Phenyl	3.2 ± 0.3	3.2 ± 0.3	Trimethylsilyl
SAX	2.3 ± 0.5	2.3 ± 0.5	None
SCX	2.8 ± 0.5	2.8 ± 0.5	None
Nitrile	3.5 ± 0.3	3.5 ± 0.3	Trimethylsilyl
Amino	3.1 ± 0.3	3.1 ± 0.3	None

ODS (C18)

REXCHROM ODS Columns

This highly retentive phase is useful for the resolution of nonpolar and moderately polar compounds. For samples that require additional retention, this phase can easily be used in ion pair chromatography. The fully endcapped REXCHROM ODS not only provides high chromatographic efficiency but also excellent sample recovery for many compounds.

Product	Particle Size	Column Length and i.d.	Product #	U.S. Price
ODS				
(Little Champ I	l) 3 μm, 100Å	$5 \text{ cm} \times 4.6 \text{ mm i.d.}$	<i>7</i> 31421	\$275.00
ODS	3 μm, 100Å	10 cm x 4.6 mm i.d.	<i>7</i> 2 <i>7</i> 118	\$300.00
ODS	5 μm, 100Å	15 cm x 4.6 mm i.d.	<i>7</i> 28118	\$300.00
ODS	5 μm, 100Å	$25 \text{ cm} \times 4.6 \text{ mm i.d.}$	728218	\$350.00
ODS	5 μm, 100Å	$25~\text{cm} \times 10.0~\text{mm}$ i.d.	728418	\$925.00

Separation of Penicillins

Column: REXCHROM ODS, 5 μm, 100Å

25 cm x 4.6 mm i.d.

Mobile Phase: (77/23) 0.02 M potassium

phosphate buffer, pH 6.8/

acetonitrile

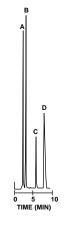
 $\begin{tabular}{lll} Flow Rate: & 1.0 mL/min \\ Load: & 10 \mu L \\ \hline \begin{tabular}{lll} Detection: & UV 254 nm \\ \hline \end{tabular}$

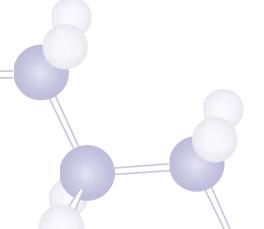
Peak Identities:

B. Ampicillin C. Benzyl Penicillin

D. Penicillin V

A. Amoxicillin





Octyl (C8)

REXCHROM Octyl Columns

Similar to the highly popular ODS (C18) phase, this monomeric octyl stationary phase can easily resolve both low and moderately polar compounds without compromising column efficiency. Ideal for analyses that require rapid elution time, the fully endcapped octyl stationary phase is often used for herbicide, pharmaceutical, and amino acid applications.

	H₃C	CH₃	
\\\\	\\``	°'_o—	SiO ₂

Product	Particle Size	Column Length and i.d.	Product #	U.S. Price
Octyl	3 μm, 100Å	10 cm x 4.6 mm i.d.	727108	\$300.00
Octyl	5 μm, 100Å	$15 \text{ cm} \times 4.6 \text{ mm} \text{ i.d.}$	<i>7</i> 28108	\$300.00
Octyl	5 μm, 100Å	$25 \text{ cm} \times 4.6 \text{ mm i.d.}$	728208	\$350.00
Octyl	5 μm, 100Å	25 cm x 10.0 mm i.d.	728408	\$925.00

Separation of Carboxylic Acids

Column: REXCHROM Octyl (C8), 5 μm, 100Å

25 cm x 4.6 mm i.d.

Mobile Phase: 0.1 M phosphoric acid

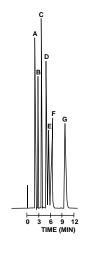
 Flow Rate:
 1.5 mL/min

 Load:
 10 μL

 Detection:
 UV 200 nm

 Peak Identities:
 A. Tartaric acid

B. Malic acid
C. Acetic acid
D. Maleic acid
E. Citric acid
F. Succinic acid
G. Propionic acid



Phenyl

REXCHROM Phenyl Columns

The phenyl bonded phases provide $\pi-\pi$ interaction between the stationary phase and the analyte. This interaction permits excellent resolution and selectivity for polar aromatic compounds, such as chlorophenol and nitroanilines.

1	Column:
H_3C Si O SiO_2	Mobile Ph Flow Rate: Load: Detection: Peak Iden:
	Product

Product	Particle Size	Column Length and i.d.	Product #	U.S. Price
Phenyl	3 μm, 100Å	10 cm x 4.6 mm i.d.	727107	\$300.00
Phenyl	5 μm, 100Å	$15 \text{ cm} \times 4.6 \text{ mm} \text{ i.d.}$	728107	\$300.00
Phenyl	5 μm, 100Å	$25 \text{ cm} \times 4.6 \text{ mm i.d.}$	728207	\$350.00
Phenyl	5 μm, 100Å	25 cm x 10.0 mm i.d.	728407	\$925.00

Separation of Aromatics

Column: REXCHROM Phenyl, 5 μm, 100Å

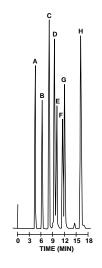
25 cm x 4.6 mm i.d.

Mobile Phase: (73/27) acetonitrile/water

low Rate: 1.0 mL/min
oad: 10 μL
Detection: UV 254 nm
leak Identities: A. Benzene

B. NaphthaleneC. BiphenylD PhenanthreneE. AnthraceneF. Fluoranthrene

G. Pyrene H. Chrysene



SAX

REXCHROM SAX Columns

The quaternary ammonium functionality of this strong anion exchange (SAX) phase maintains a constant positive charge, thus allowing retention and separation of anionic analytes, such as oligonucleotides.

(H ₃ C) ₃ N	Si SiO ₂
~ \	

Product	Particle Size	Column Length and i.d.	Product #	U.S. Price
SAX	3 μm, 100Å	10 cm x 4.6 mm i.d.	727120	\$300.00
SAX	5 μm, 100Å	$15 \text{ cm} \times 4.6 \text{ mm i.d.}$	<i>7</i> 28120	\$300.00
SAX	5 μm, 100Å	25 cm x 4.6 mm i.d.	<i>7</i> 28220	\$350.00
SAX	5 μm, 100Å	25 cm x 10.0 mm i.d.	728420	\$925.00

Separation of a Base Mixture

Column: REXCHROM SAX, 5 μm, 100Å

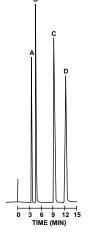
25 cm x 4.6 mm i.d.

Mobile Phase: 0.05 M sodium dihydrogen phosphate

buffer, pH 3.0

Flow Rate: 1.0 mL/minLoad: $10 \text{ }\mu\text{L}$ Detection: UV 254 nm
Peak Identities: A. Cytidine

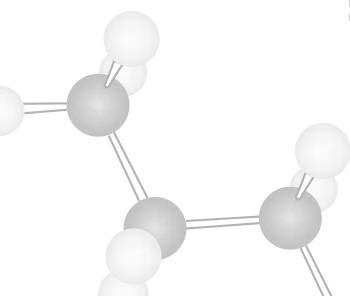
B. AdenosineC. UridineD. Guanosine



SCX

REXCHROM SCX Columns

The strong cation exchange (SCX) phase relies on the negatively charged sulfonic acid functionality for separation of catecholamines, prostaglandins, and amino acids.



Product	Particle Size	Column Length and i.d.	Product #	U.S. Price
SCX	3 μm, 100Å	$10 \text{ cm} \times 4.6 \text{ mm} \text{ i.d.}$	727130	\$300.00
SCX	5 μm, 100Å	$15 \text{ cm} \times 4.6 \text{ mm i.d.}$	<i>7</i> 28130	\$300.00
SCX	5 μm, 100Å	25 cm x 4.6 mm i.d.	728230	\$350.00
SCX	5 μm, 100Å	25 cm x 10.0 mm i.d.	728430	\$925.00

Separation of a Nucleotide Mixture

Column: REXCHROM SCX, 5 μm, 100Å

25 cm x 4.6 mm i.d.

Mobile Phase: (95/5) 0.2 M ammonium dihydrogen phosphate

buffer, pH 3.0/acetonitrile

 Flow Rate:
 1.0 mL/min

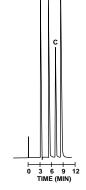
 Load:
 10 μL

 Detection:
 UV 254 nm

 Peak Identities:
 A. Uracil

 B. Guanine

C. Cytosine
D. Adenine



Silica

REXCHROM Silica Columns

Most closely resembling the phases used in thin layer chromatography (TLC), silica columns easily separate compounds containing polar functional groups, such as steroids and sterols.

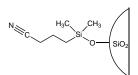
OH-	SiO ₂

Product	Particle Size	Column Length and i.d.	Product #	U.S. Price
Silica	5 μm, 100Å	$15 \text{ cm} \times 4.6 \text{ mm} \text{ i.d.}$	728100	\$280.00
Silica	5 μm, 100Å	25 cm x 4.6 mm i.d.	728200	\$325.00
Silica	5 μm, 100Å	$25~\text{cm} \times 10.0~\text{mm}$ i.d.	728400	\$825.00

Nitrile

REXCHROM Nitrile Columns

When used in normal-phase applications, nitrile bonded phases are often preferred over silica. They are less polar than silica, exhibit different selectivity and yield faster separations. Nitrile phases, in reversed-phase modes, behave as short chain alkyls with increased selectivity.



Product	Particle Size	Column Length and i.d.	Product #	U.S. Price
Nitrile	3 μm, 100Å	$10 \text{ cm} \times 4.6 \text{ mm} \text{ i.d.}$	727105	\$300.00
Nitrile	5 μm, 100Å	15 cm x 4.6 mm i.d.	728105	\$300.00
Nitrile	5 μm, 100Å	25 cm x 4.6 mm i.d.	728205	\$350.00
Nitrile	5 μm, 100Å	25 cm x 10.0 mm i.d.	728405	\$925.00

Separation of Anticonvulsants

Column: REXCHROM Nitrile, 5 μm, 100Å

25 cm x 4.6 mm i.d.

Mobile Phase: (78/22) 0.01 M sodium dihydrogen phosphate

buffer/acetonitrile

Flow Rate: 1.5 mL/min
Load: 10 µL
Detection: UV 254 nm
Peak Identities: A. Ethosuximide

B. PrimidoneC. PhenobarbitalD. Carbamazepine

E. Diphenylhydantoin



Amino

REXCHROM Amino Columns

This amino phase possesses unique characteristics that allow it to be used in normal-phase, reversed-phase, and weak anion exchange applications. In reversed-phase conditions, this phase can resolve samples containing polyaromatic hydrocarbons. However, it should not be used with amine-reactive solutes such as aldehydes, ketones, or peroxides.

$$H_2N$$
 SiO_2

Product	Particle Size	Column Length and i.d.	Product #	U.S. Price
Amino	3 μm, 100Å	$10 \text{ cm} \times 4.6 \text{ mm} \text{ i.d.}$	727103	\$300.00
Amino	5 μm, 100Å	$15 \text{ cm} \times 4.6 \text{ mm} \text{ i.d.}$	728103	\$300.00
Amino	5 μm, 100Å	25 cm x 4.6 mm i.d.	<i>7</i> 28203	\$350.00
Amino	5 μm, 100Å	25 cm x 10.0 mm i.d.	728403	\$925.00

Separation of Sugars

Column: REXCHROM Amino, 5 μm, 100Å

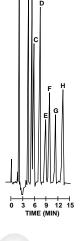
25 cm x 4.6 mm i.d.

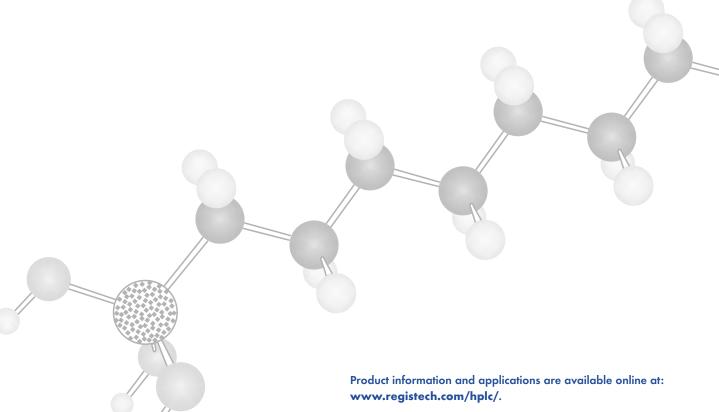
Mobile Phase: (72/28) acetonitrile/water

 $\begin{tabular}{lll} Flow Rate: & 1.5 mL/min \\ Load: & 10 ~\mu L \\ \hline \end{tabular}$ $\begin{tabular}{lll} Detection: & UV 192 nm \\ \hline \end{tabular}$ $\begin{tabular}{lll} Peak Identities: & A. Xylose \\ \hline \end{tabular}$

B. Fructose C. Glucose

D. Sucrose E. Maltose F. Lactose G. Melezitose H. Raffinose





REXCHROM REVERSIBLE UTILITY COLUMN SERIES

Little Champ II

Little Champ II Columns

As the first commercially available 5 cm, 3 μ m ODS column, the Regis Little Champ became well accepted for its rapid, efficient, and effective separations. No other product could match the versatility of this column.

Like its predecessor (the Little Champ), the Little Champ II is a popular choice with chromatographers. Starting with 3 μ m silica and improved ODS bonding chemistry, this multifaceted column repeatedly outperforms larger columns and yields clean separations in minimal time. The Little Champ II has demonstrated application excellence in:

- Process monitoring
- Routine analysis
- Enzyme separations
- · Combinatorial chemistry

Recently, the Little Champ II has proven to be effective in combinatorial chemistry, where it has been used in automated high throughput separations. These separations are easily scaled up to the larger 10 mm i.d. preparative version of this column.

Product	Particle Size	Column Length and i.d.	Product #	U.S. Price
Little				
Champ II	3 μm, 100Å	$5~\mathrm{cm} \times 4.6~\mathrm{mm}$ i.d.	<i>7</i> 31421	\$275.00

Rapid Separation of Catecholamines

Column: REXCHROM Little Champ II (ODS), 3 μm, 100Å

5 cm x 4.6 mm i.d.

Mobile Phase: (82/18) 0.005 M S8 ion pair concentrates/

methanol, pH 2.5

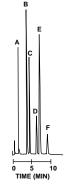
 $\begin{tabular}{llll} Flow Rate: & 1.0 mL/min \\ Load: & 10 μL \\ \\ Detection: & UV 280 nm \\ \end{tabular}$

Peak Identities: A. 3,4-Dihydroxyphenylacetic acid

B. NorepinephrineC. Epinephrine

D. 3,4-Dihydroxybenzylamine E. 3,4-Dihydroxyphenylalanine

F. Dopamine



Val-U-Pak HP

Val-U-Pak HP Columns

Regis offers the Val-U-Pak HP in an effort to provide a high performance and highly efficient column at the lowest possible price. This column contains spherical, 5 μ m, 100Å, fully endcapped ODS silica. The Val-U-Pak HP column is batch tested, eliminating the expense of individual testing. The low cost of a Val-U-Pak HP permits the dedication of a column to a specific application.

Product	Particle Size	Column Length and i.d.	Product #	U.S. Price
Val-U-Pak HP	5 μm, 100Å	$25 \text{ cm} \times 4.6 \text{ mm i.d.}$	<i>7</i> 31901	\$300.00
Val-U-Pak HP	5 μm, 100Å	$25~\text{cm} \times 10.0~\text{mm}$ i.d.	<i>7</i> 31903	\$825.00

Separation of Saccharine and Alkaloids

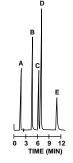
Column: REXCHROM Val-U-Pak HP (ODS), 5 μm, 100Å

25 cm x 4.6 mm i.d.

Mobile Phase: (75/25) methanol/water

 $\begin{array}{lll} \textbf{Flow Rate:} & 1.0 \text{ mL/min} \\ \textbf{Load:} & 10 \text{ }\mu\text{L} \\ \textbf{Detection:} & \text{UV 254 nm} \\ \textbf{Peak Identities:} & \text{A. Saccharine} \\ \text{B. Theobromine} \\ \end{array}$

C. Dyphylline
D. Theophylline
E. Caffeine



REXCHROM REVERSIBLE UTILITY COLUMN SERIES

Base Columns

REXCHROM Base Columns

In the past, the separation of basic compounds by HPLC has been generally unsuccessful. Base columns, developed specifically for this class of compounds, often have low efficiencies which, in turn, adversely affect both resolution and peak shape. The primary cause of this low efficiency is residual acidic sites on the silica, which interact with the basic analytes. Often the modification of mobile phase concentration and components is not a sufficient remedy.

REXCHROM base columns, available in both octyl and ODS bonded phases, virtually eliminate the problem of residual acidic sites. Diligent monitoring during the base deactivation procedure results in the complete masking of these acidic sites. With this new generation of columns, separations can be performed easily with mobile phases of either aqueous methanol or acetonitrile - no buffers are necessary.

Product	Particle Size	Column Length and i.d.	Product #	U.S. Price
ODS	5 μm, 100Å	$15 \text{ cm} \times 4.6 \text{ mm i.d.}$	<i>7</i> 28618	\$380.00
ODS	5 μm, 100Å	25 cm x 4.6 mm i.d	728718	\$400.00
Octyl	5 μm, 100Å	15 cm x 4.6 mm i.d	728608	\$380.00
Octyl	5 μm, 100Å	25 cm x 4.6 mm i.d	728708	\$400.00

Separation of a Well-Known Tailing Mixture

REXCHROM BASE ODS, 5 μm, 100Å Column:

 $25 \text{ cm} \times 4.6 \text{ mm} \text{ i.d.}$

Mobile Phase: (85/15) acetonitrile/water

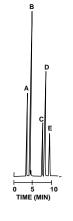
Flow Rate: 1.0 mL/min Load: 10 µL Detection: UV 254 nm

Peak Identities: A. 5-Phenyl-1-pentanol

B. N,N-Diethylaniline C. 1-Phenylhexane D. 2,6-Di-tert-butylpyridine E. 1-Phenylheptane

Reference: Kohler, J.; Kirkland, J. J.; J.Chromatog. 1987,

385, 125-150.

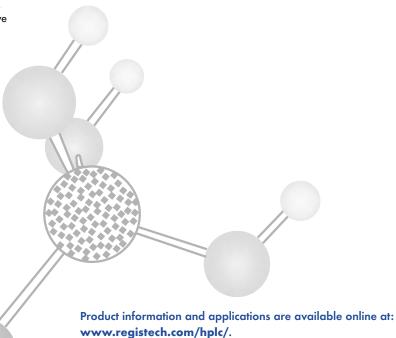


Workhorse II

Workhorse II ODS Column

For separations in which longer column length and low back pressures are essential, Regis offers the ODS Workhorse II column. This 30 cm column containing spherical 10 µm particles allows both higher flow rates and larger sample volumes. Designed for day-in, day-out routine analysis, the Workhorse II has proven to be an effective tool in drug formulation and protein studies.

Product	Particle Size	Column Length and i.d.	Product #	U.S. Price
ODS	10 μm, 100Å	$30 \text{ cm} \times 4.6 \text{ mm i.d.}$	<i>7</i> 31128	\$350.00



13

REXCHROM GUARD COLUMNS

Guard Columns

REXCHROM Guard Columns

The routine use of a guard column will significantly extend the life of any analytical column. Ideally, guard columns need to retain irreversibly adsorbed materials and be chromatographically efficient.

Regis offers a wide variety of 3 μm and 5 μm guard cartridges, conveniently packaged either in a kit or replacement set. The kit is comprised of two disposable guard cartridges, a reusable holder, and a column coupler. The replacement set contains three guard cartridges.

The guard cartridges, which contain spherical material, are packed in 1 cm x 3.0 mm i.d. columns. Proficient packing, like that of the analytical columns, allows the guard column to be used in either flow direction.

Product	Particle Size	Column Length and i.d.	Product #	U.S. Price
ODS (C18) Guard Kit*	3 μm, 100Å	$1.0~\text{cm} \times 3.0~\text{mm}$ i.d.	721184	\$225.00
ODS (C18) Guard Kit re	eplacements**		<i>7</i> 21185	\$130.00
ODS (C18) Guard Kit*	5 μm, 100Å	$1.0~\text{cm} \times 3.0~\text{mm}$ i.d.	<i>7</i> 411 <i>7</i> 4	\$200.00
ODS (C18) Guard Kit re	eplacements**		<i>7</i> 411 <i>75</i>	\$120.00
Octyl (C8) Guard Kit*	3 μm, 100Å	1.0 cm x 3.0 mm i.d.	721084	\$225.00
Octyl (C8) Guard Kit re	placements**		721085	\$130.00
Octyl (C8) Guard Kit*	5 μm, 100Å	1.0 cm x 3.0 mm i.d.	<i>7</i> 41084	\$200.00
Octyl (C8) Guard Kit re	placements**		<i>7</i> 41085	\$120.00
Phenyl Guard Kit*	3 μm, 100Å	1.0 cm x 3.0 mm i.d.	721074	\$225.00
Phenyl Guard Kit replac			721075	\$130.00
Phenyl Guard Kit*	5 μm, 100Å	1.0 cm x 3.0 mm i.d.	<i>7</i> 410 <i>7</i> 4	\$200.00
Phenyl Guard Kit replac	ements**		741075	\$120.00
SAX Guard Kit*	3 μm, 100Å	1.0 cm x 3.0 mm i.d.	721304	\$225.00
SAX Guard Kit replacem			721305	\$130.00
SAX Guard Kit*	5 μm, 100Å	1.0 cm x 3.0 mm i.d.	<i>7</i> 41304	\$200.00
SAX Guard Kit replacem	ents**		<i>7</i> 41305	\$120.00
SCX Guard Kit*	3 μm, 100Å	1.0 cm x 3.0 mm i.d.	721334	\$225.00
SCX Guard Kit replacem			721335	\$130.00
SCX Guard Kit*	5 μm, 100Å	1.0 cm x 3.0 mm i.d.	<i>7</i> 41314	\$200.00
SCX Guard Kit replacem	nents**		<i>7</i> 41315	\$120.00
	•			
Nitrile Guard Kit*	3 μm, 100Å	1.0 cm x 3.0 mm i.d.	721054	\$225.00
Nitrile Guard Kit replace			721055	\$130.00
Nitrile Guard Kit*	5 μm, 100Å	1.0 cm x 3.0 mm i.d.	<i>7</i> 41054	\$200.00
Nitrile Guard Kit replace	ements**		<i>7</i> 41055	\$120.00
Amino Guard Kit*	3 μm, 100Å	1.0 cm x 3.0 mm i.d.	721034	\$225.00
Amino Guard Kit replace			721035	\$130.00
Amino Guard Kit*	5 μm, 100Å	1.0 cm x 3.0 mm i.d.	741034	\$200.00
Amino Guard Kit replace	ements**		<i>7</i> 41035	\$120.00

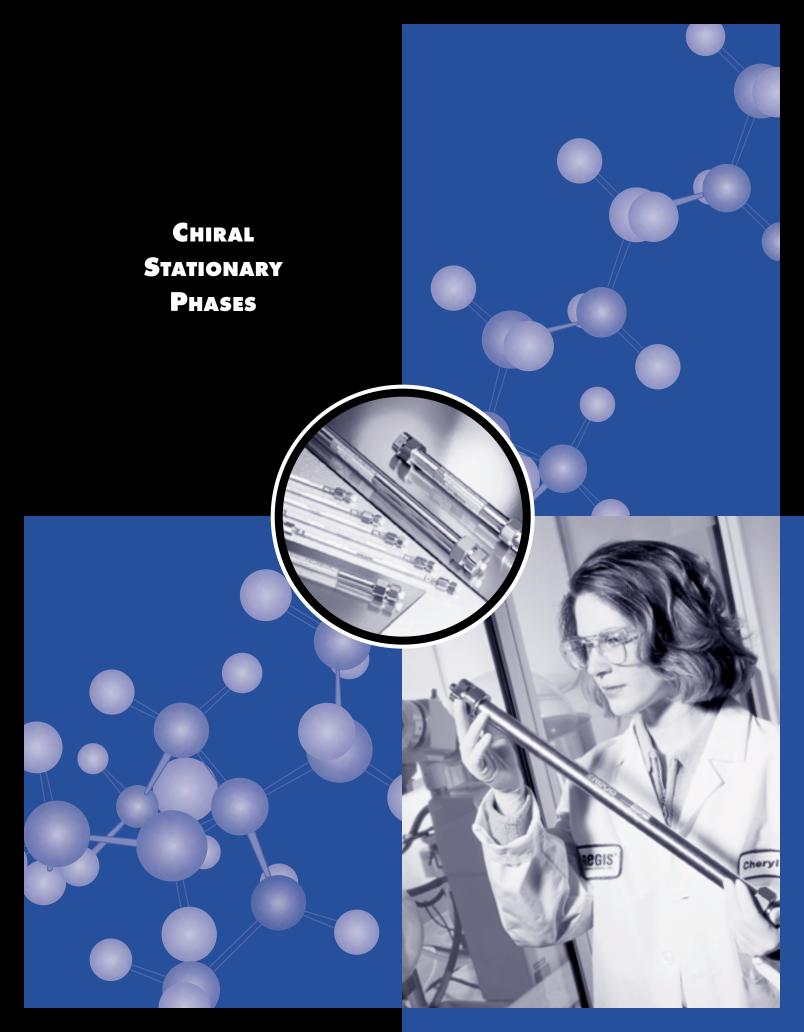


^{**} Includes: 3 guard cartridges

For additional information on	
Conventional HPLC Columns, check	
our Web site at www.registech.com/hplc/	
or contact Regis at:	

(800) 323-8144 ext. 649 (847) 967-6000 ext. 649 e-mail us at: sales@registech.com.

Product Accessories	Product #	U.S. Price
Replacement guard cartridge holder		
for analytical columns	731441	\$125.00
Internal fitting coupler		
for analytical columns	731443	\$35.00



CHIRAL STATIONARY PHASES

Chirality has become vitally important in the pharmaceutical, chemical and agricultural industries.

The differences which make

compounds chiral can produce critically different pharmacological effects in biological systems.

As a result, demand for stereoselective separation techniques and analytical assays to evaluate the enantiomeric purity of chiral compounds has increased.

Chiral chromatography has become a necessary tool—not only for the analytical determination of enantiomeric purity, but also for the isolation of pure enantiomers.

Chiral Chromatography

Regis Technologies is proud to be a leader in chiral separations and serve both the analytical and preparative needs of chromatographers and researchers worldwide. Regis offers three different classes of Chiral Stationary Phases (CSPs):

- Pirkle-Concept
- Davankov Ligand Exchange
- Protein-based

Regis manufactures a complete line of Pirkle
Chiral Stationary Phases and Davankov
Ligand Exchange columns at its pharmaceutical
manufacturing facility. Columns range from
analytical to preparative in size. A line of
protein-based chiral stationary phases is also
available. All products meet rigorous manufacturing
and quality control specifications before release.



Pirkle Stationary Phases

In 1980, Regis Technologies, along with Professor William Pirkle, of the University of Illinois, introduced the Pirkle Chiral Stationary Phases. These Chiral Stationary Phases offer many advantages:

- Enantiomer separation on a wide variety of compound groups
- · Column durability resulting from covalent phase bonding
- · Ability to invert elution order
- Availability of analytical- to preparative-sized columns and bulk packing material
- Universal solvent compatibility

Enantiomer Separation

Regis manufactures ten Pirkle CSPs. These can separate a wide variety of enantiomers in numerous compound groups. Examples include:

- Aryl Propionic Acid Non-Sterodial Anti-Inflammatory Drugs (NSAIDs)
- Agricultural Compounds
- Natural Products
- β-Blockers
- · Many Pharmaceuticals



Additional examples of enantiomer separations can be found in the Regis Chiral Application Guide IV or on our Web site at www.registech.com/chiral/. Our Web site is updated monthly with new applications and current chiral events.

Column Durability

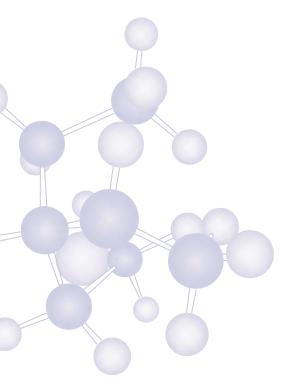
The Pirkle Chiral Stationary Phases are covalently bonded to the silica, providing excellent column durability. Covalently bonded phases assure long-lasting columns and offer added benefits for preparative columns. Covalently bonded preparative columns are longer lasting than their coated, preparative column counterparts because with use, noncovalent coatings can leach off. Additional benefits include the columns' capacity to tolerate sample overload.

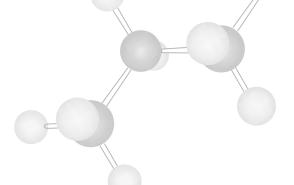
Ability to Invert Elution Order

An important advantage of the Pirkle Chiral Stationary Phases is the ability to invert elution order by using the same type of CSP, but with the opposite absolute configuration. As a result, it is possible to have the trace enantiomer elute before the major — a desirable feature for enantiomeric purity determinations. For preparative separations it is beneficial to elute the desired component first.

Analytical and Preparative-Sized Columns

All of Regis' Pirkle HPLC columns are available in both analytical and preparative sizes. Since all chiral stationary phases are manufactured on-site, Regis can pack special or custom-sized columns quickly and easily.





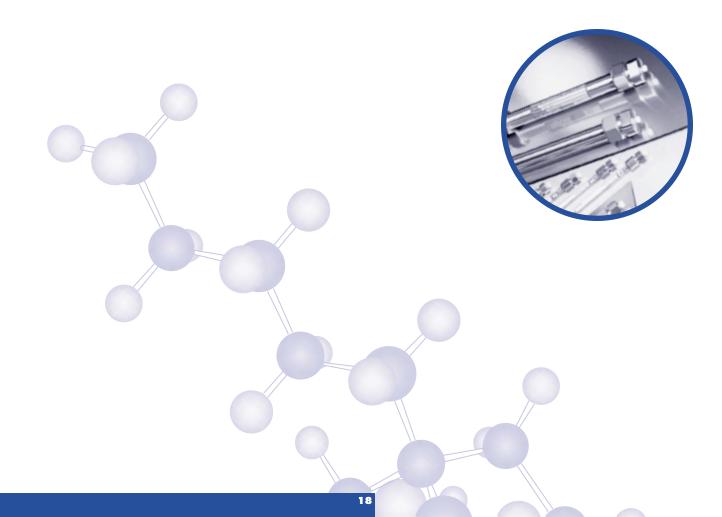




Choice of mobile phase is not a limitation with the Pirkle HPLC columns. They are compatible with most mobile phases. The pH of the mobile phase, however, must be between 2.5 and 7.5. Both normal-phase and reversed-phase modes can be used, although normal-phase is most common. For normal-phase separations, the classic mobile phase is a binary or ternary mixture of a hydrocarbon and a modifier, usually an aliphatic alcohol.

Typical uncharged organic modifers include ethanol, isopropanol and butanol. Under reversed-phase conditions, water-alcohol mixtures or aqueous phosphate buffers with charged organic modifiers are also employed.

Super and subcritical (SFC and SubFC) fluid chromatography, utilizing carbon dioxide, has also been introduced as a promising technique for the separation of enantiomers using Pirkle Chiral Stationary Phases.



Whelk-O® 1

Analytical to Preparative Columns

The Whelk-O 1 is useful for the separation of underivatized enantiomers in a number of families including amides, epoxides, esters, ureas, carbamates, ethers, aziridines, phosphonates, aldehydes, ketones, carboxylic acids, alcohols and non-steroidal anti-inflammatory drugs (NSAIDs).

This π -electron acceptor/ π -electron donor phase exhibits an extraordinary degree of generality. The broad versatility observed on the Whelk-O 1 column compares favorably with polysaccharide-derived chiral stationary phases.

In addition, because Whelk-O 1 is covalently bonded to the support, the phase is compatible with all commonly used mobile phases, including aqueous systems — a distinct advantage over polysaccharidederived chiral stationary phases. Other advantages include column durability, excellent efficiency, ability to invert elution order and excellent preparative capacity.

Product	Particle Size	Column Length and i.d.	Product #	U.S. Price
Spherical silica:				
(R,R)-Whelk-O 1	5 μm, 100Å	25 cm x 4.6 mm i.d.	786201	\$1,500.00
(R,R)-Whelk-O 1	5 μm, 100Å	$25~\text{cm} \times 10.0~\text{mm}$ i.d.	786202	\$5,000.00
(S,S)-Whelk-O 1	5 μm, 100Å	25 cm x 4.6 mm i.d.	<i>7</i> 86101	\$1,500.00
(S,S)-Whelk-O 1	5 μm, 100Å	$25~\text{cm} \times 10.0~\text{mm}$ i.d.	786102	\$5,000.00
Spherical Kromas	il silica:			
(R,R)-Whelk-O 1	10 μm, 100Å	25 cm x 4.6 mm i.d.	786515	\$1,500.00
(R,R)-Whelk-O 1	10 μm, 100Å	$25~\text{cm} \times 10.0~\text{mm}$ i.d.	786525	\$5,000.00
(R,R)-Whelk-O 1	10 μm, 100Å	25 cm x 21.1 mm i.d.	786535	\$11,000.00
(R,R)-Whelk-O 1	10 μm, 100Å	50 cm x 21.1 mm i.d.	786545	\$18,000.00
(S,S)-Whelk-O 1	10 μm, 100Å	25 cm x 4.6 mm i.d.	786615	\$1,500.00
(S,S)-Whelk-O 1	10 μm, 100Å	$25~\text{cm} \times 10.0~\text{mm}$ i.d.	786625	\$5,000.00
(S,S)-Whelk-O 1	10 μm, 100Å	25 cm x 21.1 mm i.d.	786635	\$11,000.00
(S,S)-Whelk-O 1	10 μm, 100Å	50 cm x 21.1 mm i.d.	786645	\$18,000.00

Ibuprofen

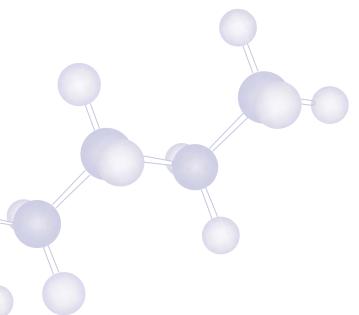
Column: Whelk-O 1

25 cm x 4.6 mm i.d

Mobile Phase: (98/2/0.05)

hexane/isopropanol/acetic acid

Flow Rate: 0.9 mL/min
Load: 20 µL
Detection: UV 254 nm
Run Time: 8 min
Reference: 2



Whelk-O is a registered trademark of Regis Technologies, Inc.

Whelk-0® 2Analytical to Preparative Columns

Our newest addition to the Whelk-O line of chiral stationary phases is the Whelk-O 2. The Whelk-O 2 is the covalent trifunctional version of the Whelk-O 1. The Whelk-O 2 retains the same chiral selector but incorporates a trifunctional linkage to the silica support. In most cases, the enantioselectivity remains the same as that obtained with the Whelk-O 1.

Whelk-O 2 was designed to improve the resistance of the stationary phase to hydrolysis while using strong organic modifiers such as trifluoroacetic acid. The Whelk-O 2 is ideal for preparative separations since the material is bonded on 10 μm , 100Å spherical Kromasil silica. This allows the preparative chromatographer to perform method development on an analytical column and immediately scale up to larger diameter columns.

Particle Size	Column Length and i.d.	Product #	U.S. Price
silica:			
10 μm, 100Å	$25 \text{ cm} \times 4.6 \text{ mm i.d.}$	786315	\$1,600.00
10 μm, 100Å	$25~\text{cm} \times 10.0~\text{mm}$ i.d.	786325	\$5,000.00
10 μm, 100Å	$25 \text{ cm} \times 21.1 \text{ mm i.d.}$	786335	\$11,000.00
10 μm, 100Å	50 cm x 21.1 mm i.d.	786345	\$18,000.00
10 μm, 100Å	$25 \text{ cm} \times 4.6 \text{ mm i.d.}$	786415	\$1,600.00
10 μm, 100Å	$25~\text{cm} \times 10.0~\text{mm}$ i.d.	786425	\$5,000.00
10 μm, 100Å	25 cm x 21.1 mm i.d.	786435	\$11,000.00
10 μm, 100Å	50 cm x 21.1 mm i.d.	786445	\$18,000.00
	silica: 10 μm, 100Å	silica: 10 μm, 100Å 25 cm x 4.6 mm i.d. 10 μm, 100Å 25 cm x 10.0 mm i.d. 10 μm, 100Å 25 cm x 21.1 mm i.d. 10 μm, 100Å 50 cm x 21.1 mm i.d. 10 μm, 100Å 25 cm x 4.6 mm i.d. 10 μm, 100Å 25 cm x 10.0 mm i.d. 10 μm, 100Å 25 cm x 21.1 mm i.d.	10 μm, 100Å 25 cm x 4.6 mm i.d. 786315 10 μm, 100Å 25 cm x 10.0 mm i.d. 786325 10 μm, 100Å 25 cm x 21.1 mm i.d. 786335 10 μm, 100Å 50 cm x 21.1 mm i.d. 786345 10 μm, 100Å 25 cm x 4.6 mm i.d. 786415 10 μm, 100Å 25 cm x 10.0 mm i.d. 786425 10 μm, 100Å 25 cm x 21.1 mm i.d. 786435

Leucine

Analytical and Semi-Preparative Columns

The π -acceptor leucine CSP is based on 3,5-dinitrobenzoyl leucine, covalently bonded to 5 μ m aminopropyl silica. Columns derived from either L- or D-leucine are available. This phase demonstrates enhanced enantioselectivities for several classes of compounds, including benzodiazapines.

Product	Particle Size	Column Length and i.d.	Product #	U.S. Price
D-Leucine	5 μm, 100Å	$25 \text{ cm} \times 4.6 \text{ mm} \text{ i.d.}$	731054	\$800.00
D-Leucine	5 μm, 100Å	25 cm x 10.0 mm i.d.	731254	\$1,800.00
L-Leucine	5 μm, 100Å	25 cm x 4.6 mm i.d.	731041	\$ <i>75</i> 0.00
L-Leucine	5 μm, 100Å	25 cm x 10.0 mm i.d.	<i>7</i> 31241	\$1,600.00

Hexobarbital Column: L-Leucine $25 \text{ cm} \times 4.6 \text{ mm} \text{ i.d.}$ **Mobile Phase:** (95/5) hexane/ethanol Flow Rate: 0.7 mL/minLoad: 0.686 mg/mL UV 254 nm Detection: **Run Time:** 16 min k'1: 2.89 1.10 α:

O N OF

Whelk-O is a registered trademark of Regis Technologies, Inc.

Phenylglycine

Analytical and Semi-Preparative Columns

Phenylglycine, a π -acceptor chiral phase, is based on 3,5-dinitrobenzoyl phenylglycine, covalently bonded to 5 μ m aminopropyl silica. Phenylglycine columns are available in both L- and D- configurations.

This CSP resolves a wide variety of compounds containing π -basic groups, including: aryl-substituted cyclic sulfoxides, bi- β -naphthol and its analogs, α -indanol and α -tetralol analogs, and aryl-substituted hydantoins.

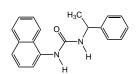
Product	Particle Size	Column Length and i.d.	Product #	U.S. Price
D-Phenylglycine	5 μm, 100Å	$25 \text{ cm} \times 4.6 \text{ mm} \text{ i.d.}$	<i>7</i> 31021	\$700.00
D-Phenylglycine	5 μm, 100Å	25 cm x 10.0 mm i.d.	731221	\$1,400.00
L-Phenylglycine	5 μm, 100Å	25 cm x 4.6 mm i.d.	731024	\$700.00
L-Phenylglycine	5 μm, 100Å	25 cm x 10.0 mm i.d.	731224	\$1,400.00

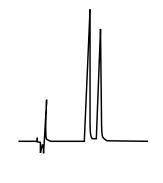
N-(1-Naphthyl)-N'-(1-methylbenzyl) urea

Column: D-Phenylglycine

25 cm x 4.6 mm i.d.

Mobile Phase: (70/30) hexane/ethanol





β**-Gem 1**Analytical and Semi-Preparative Columns

β-Gem 1 is a π-acceptor chiral stationary phase and is prepared by covalently bonding N-3,5-dinitrobenzoyl-3-amino-3-phenyl-2-{1,1-dimethylethyl}-propanoate, to 5 μ m silica through an ester linkage.

In many cases, this chiral phase considerably outperforms its widely used analog, phenylglycine. It can separate anilide derivatives of chiral carboxylic acids, including nonsteroidal anti-inflammatory agents.

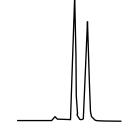
Product	Particle Size	Column Length and i.d.	Product #	U.S. Price
(R,R)-β-GEM 1	5 μm, 100Å	$25 \text{ cm} \times 4.6 \text{ mm} \text{ i.d.}$	731043	\$1,400.00
(R,R)-β-GEM 1	5 μm, 100Å	$25~\text{cm} \times 10.0~\text{mm}$ i.d.	731243	\$4,400.00
(S,S)-β-GEM 1	5 μm, 100Å	25 cm x 4.6 mm i.d.	<i>7</i> 31029	\$1,400.00
(S,S)-β-GEM 1	5 μm, 100Å	$25~\text{cm} \times 10.0~\text{mm}$ i.d.	731229	\$4,400.00

trans-(R)7,8-Dihydroxy-9,10-epoxy-7,8,9,10-tetrahydrobenzo[a]pyrene

Column: (*R,R*)-β-Gem 1

25 cm x 4.6 mm i.d.

Mobile Phase: (60/40) hexane/ethanol



\alpha\text{-Burke 2}Analytical and Semi-Preparative Columns

The α -Burke 2 phase is derived from dimethyl N-3,5-dinitro-benzoyl- α -amino-2,2-dimethyl-4-pentenyl phosphonate covalently bound to 5 μ m silica. This π -acceptor chiral stationary phase is particularly valuable in the HPLC separation of β -blocker enantiomers, an important class of cardiovascular drugs whose enantiomers often exhibit differing pharmacological activities. The α -Burke 2 has been specifically designed to separate the enantiomers of β -blockers without chemical derivatization. In addition, it also resolves the enantiomers of many compounds separated on π -acceptor Pirkle type chiral stationary phases.

Product	Particle Size	Column Length and i.d.	Product #	U.S. Price
(R)-α-Burke 2	5 μm, 100Å	$25 \text{ cm} \times 4.6 \text{ mm} \text{ i.d.}$	735035	\$1,400.00
(R)-α-Burke 2	5 μm, 100Å	25 cm x 10.0 mm i.d.	735235	\$4,400.00
(S)-α-Burke 2	5 μm, 100Å	25 cm x 4.6 mm i.d.	735037	\$1,400.00
(S)-α-Burke 2	5 μm, 100Å	25 cm x 10.0 mm i.d.	735237	\$4,400.00

Betaxolol

Column: α-Burke 2

25 cm x 4.6 mm i.d.

Mobile Phase: (85/10/5) CH₂C1₂/EtOH/MeOH

 Flow Rate:
 1 mL/min

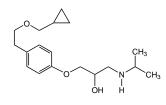
 Detection:
 UV 254 nm

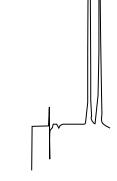
 Run Time:
 11 min

 k'1:
 2.36

 α:
 1.25

 Reference:
 4





Pirkle 1-J

Analytical and Semi-Preparative Columns

The Pirkle 1-J column is the latest in a series of CSPs from the research laboratories of Professor Pirkle. This new CSP contains an unusual β -lactam structure which significantly alters its molecular recognition properties. The Pirkle 1-J is useful for the direct separation of underivatized β -blocker enantiomers. It can also be used for the separation of the enantiomers of arylpropionic acid NSAIDs, as well as other drugs.

Product	Particle Size	Column Length and i.d.	Product #	U.S. Price
(3R, 4S)-Pirkle 1-J	5 μm, 100Å	$25 \text{ cm} \times 4.6 \text{ mm} \text{ i.d.}$	731044	\$1,400.00
(3R, 4S)-Pirkle 1-J	5 μm, 100Å	25 cm x 10.0 mm i.d.	731244	\$4,400.00
(3S, 4R)-Pirkle 1-J	5 μm, 100Å	25 cm x 4.6 mm i.d.	731045	\$1,400.00
(3S, 4R)-Pirkle 1-J	5 μm, 100Å	$25~\text{cm} \times 10.0~\text{mm}$ i.d.	731245	\$4,400.00

Pindolol

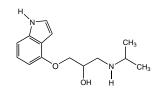
Column: Pirkle 1-J

25 cm x 4.6 mm i.d.

Mobile Phase: (80/20) methylene chloride/ethanol,

+0.04M ammonium acetate

 $\begin{tabular}{lll} Flow Rate: & 1.0 mL/min \\ Load: & 10 ~\mu L \\ \hline \end{tabular}$ $\begin{tabular}{lll} Detection: & UV 254 ~nm \\ \hline \end{tabular}$





Naphthylleucine

Analytical and Semi-Preparative Columns

The naphthylleucine phase, a π -electrondonor, is based on N-(1-naphthyl) leucine, covalently bonded to 5 μ m silica through an ester linkage.

This phase resolves DNB derivatives of amino acids as the free acid when used in reversed-phase mode. In the classic normal-phase, this CSP can resolve the amides and esters of DNB amines, alcohols and amino acids.

Product	Particle Size	Column Length and i.d.	Product #	U.S. Price
L-Naphthylleucine	5 μm, 100Å	$25 \text{ cm} \times 4.6 \text{ mm} \text{ i.d.}$	731034	\$800.00
L-Naphthylleucine	5 μm, 100Å	25 cm x 10.0 mm i.d.	731234	\$1,900.00

N-(3,5-Dinitrobenzoyl) valine

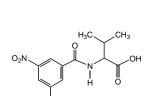
Column: Naphthylleucine

25 cm x 4.6 mm i.d.

Mobile Phase: (80/20) methanol/10mM KH₂PO₄,

pH 6.86 + 0.5 mM Q6

 $\begin{tabular}{llll} Flow Rate: & 1.0 mL/min \\ Load: & 5 \mu L \\ \hline \begin{tabular}{llll} Detection: & UV 254 nm \\ Run Time: & 4 min \\ \hline \end{tabular}$

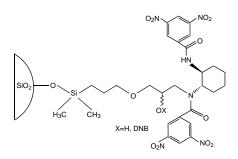




DACH-DNB

Analytical to Preparative Columns

The innovative DACH-DNB CSP was designed by Italian chemists Drs. Francesco Gasparrini, Misiti and Villani at Rome University "La Sapienza." The DACH-DNB CSP, which contains the 3,5-dinitrobenzoyl derivative of 1,2-diaminocyclohexane, has been found to resolve a broad range of racemate classes including amides, alcohols, esters, ketones, acids, sulfoxides, phosphine oxides, selenoxides, phosphonates, thiophosphineoxide, phosphineselenide, phosphine-borane, beta-lactams, organometallics, atropisomers and heterocycles.



Product	Particle Size	Column Length and i.d.	Product #	U.S. Price
(R,R)-DACH-DNB	5 μm, 100Å	$25 \text{ cm} \times 4.6 \text{ mm} \text{ i.d.}$	788101	\$1,400.00
(R,R)-DACH-DNB	5 μm, 100Å	25 cm x 10.0 mm i.d.	788102	\$4,400.00
(R,R)-DACH-DNB	10 μm, 100Å	25 cm x 21.1 mm i.d.	788103	\$11,000.00
(S,S)-DACH-DNB	5 μm, 100Å	25 cm x 4.6 mm i.d.	788201	\$1,400.00
(S,S)-DACH-DNB	5 μm, 100Å	$25~\text{cm} \times 10.0~\text{mm}$ i.d.	788202	\$4,400.00
(S,S)-DACH-DNB	10 μm, 100Å	25 cm x 21.1 mm i.d.	788203	\$11,000.00

Sulfoxide

Column: (R,R)-DACH-DNB

25 cm x 4.6 mm

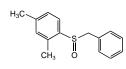
Mobile Phase: $(95/5) CH_2CI_2/IPA$

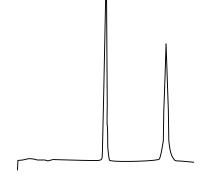
Flow Rate: 1.0 ml/min

Detection: UV 254 nm

Run Time: 15.0 min

k'1: 2.15
α: 2.05





ULMO

Analytical to Preparative Columns

The ULMO chiral stationary phase was developed by Austrian researchers Uray, Lindner and Maier. The ULMO CSP is based on a 3,5-dintrobenzoyl derivative of diphenylethylenediamine. This CSP has a general ability to separate the enantiomers of many racemate classes and is particularly good at separating the enantiomers of aryl carbinols.

Product	Particle Size	Column Length and i.d.	Product #	U.S. Price
(S,S)-ULMO	5 μm, 100Å	$25 \text{ cm} \times 4.6 \text{ mm} \text{ i.d.}$	787100	\$1,400.00
(S,S)-ULMO	5 μm, 100Å	$25~\text{cm} \times 10.0~\text{mm}$ i.d.	<i>7</i> 8 <i>7</i> 101	\$4,400.00
(S,S)-ULMO	10 μm, 100Å	$25 \text{ cm} \times 21.1 \text{ mm i.d.}$	787102	\$11,000.00
(R,R)-ULMO	5 μm, 100Å	$25 \text{ cm} \times 4.6 \text{ mm i.d.}$	787200	\$1,400.00
(R,R)-ULMO	5 μm, 100Å	$25~\text{cm} \times 10.0~\text{mm}$ i.d.	787201	\$4,400.00
(R,R)-ULMO	10 μm, 100Å	$25~\text{cm} \times 21.1~\text{mm} \text{ i.d.}$	787202	\$11,000.00

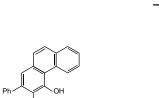
Vapol

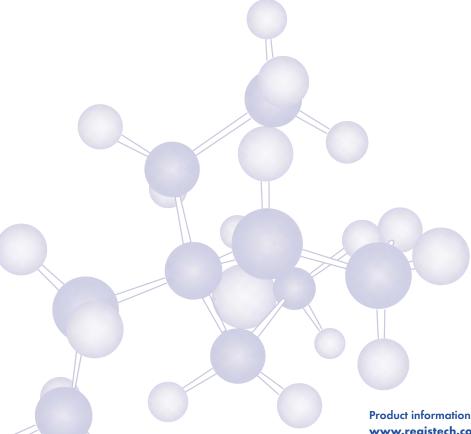
Column: (R,R)-ULMO

25 cm x 4.6 mm

Mobile Phase: 100% methanol
Flow Rate: 1.5 mL/min
Detection: UV 254 nm
Run Time: 13 min

k'₁: 1.74 α: 3.37





DAVANKOV LIGAND EXCHANGE CHIRAL STATIONARY PHASE

Davankov Ligand Exchange Chiral Stationary Phase

The Davankov chiral stationary phase is useful for the separation of underivatized amino acid enantiomers. This phase operates according to the principles of ligand-exchange chromatography (LEC), a technique pioneered by Professor V. Davankov.

The Davankov column requires a mobile phase of aqueous methanol containing copper(II) acetate. Enantioselectivity is extremely high with alphas up to 16 being reported. Regis provides either a Davankov HPLC column, or a kit which allows the user to convert a standard ODS column into a Davankov Chiral Stationary phase.

Both of these Davankov products maintain a stable coating compatible with those mobile phases generally used in amino acid separations.

Davankov Column

A pre-converted Davankov column complete with care and use guide, column test conditions and performance results is available.

Davankov Reagent A Kit

Regis provides the Davankov Reagent A kit, which contains Davankov Reagent A, a hydroxyproline derivative and copper(II) acetate (sufficient quantities to coat one 15 cm column and prepare mobile phase). The column coating procedure involves dissolving the Davankov Reagent A into methanol/water (80/20) and pumping this mixture through the column. This is followed by a wash with a concentrated solution of Cu(OAc)2 in methanol/water (15/85). Detailed coating procedures are included with the kit.

Product	Particle Size	Column Length and i.d.	Product #	U.S. Price
Davankov Column	5 μm, 100Å	$15 \text{ cm} \times 4.6 \text{ mm} \text{ i.d.}$	<i>7</i> 31653	\$700.00
Davankov Reagent A Kit			<i>7</i> 31650	\$300.00
REXCHROM ODS Column	5 μm, 100Å	15 cm x 4.6 mm i.d.	728118	\$300.00

DL-Leucine and DL-Norvaline

REXCHROM Davankov A Column:

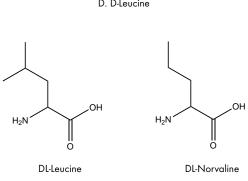
15 cm x 4.6 mm i.d.

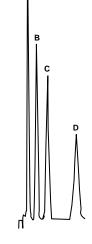
(65/35) $10^{-4} \; \mathrm{M} \; \mathrm{CuAc}_{2}, \mathrm{pH} \; 5.0/\mathrm{methanol}$ **Mobile Phase:**

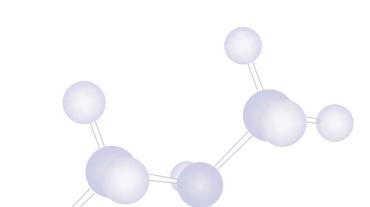
Flow rate: 2.0 mL/min **Detection:** UV 254 nm **Run Time:** 16 min Peak Identities: A. L-Norvaline

B. L-Leucine C. D-Norvaline

D. D-Leucine







PROTEIN-BASED CHIRAL STATIONARY PHASES

Protein-Based Chiral Stationary Phases

Regis carries a line of protein-based chiral columns manufactured by ChromTech AB. These include:

- Chiral AGP-(α–glycoprotein)
- Chiral CBH-(cellobiohydrolase)
- Chiral HSA-(human serum albumin)

For additional product information and a Protein-Based Stationary Phase Application Guide, please contact Regis at sales@registech.com.

Chiral AGPMicro, Analytical and Semi-Preparative Columns

Chiral AGP is the second generation chiral selector based on the α_1 -acid glycoprotein (α_1 -AGP) as the chiral stationary phase. The AGP has been immobilized on spherical, 5 μm particles. The Chiral AGP column is typically used in the reversed-phase mode, where it can be used for the resolution of an extremely broad range of chiral compounds, such as amines, (primary, secondary, tertiary and quaternary ammonium), acids, esters, sulphoxides, amides, and alcohols. The enantioselectivity and the retention can easily be regulated by the pH of the mobile phase, the buffer concentration and the nature and concentration of the organic modifier.

Product	Particle Size	Column Length and i.d.	Product #	U.S. Price
Chiral AGP	5 μm	$15 \text{ cm} \times 0.18 \text{ mm} \text{ i.d.}$	732195	\$1,600.00
Chiral AGP	5 μm	10 cm x 1 mm i.d.	732194	\$1,950.00
Chiral AGP	5 μm	$10 \text{ cm} \times 2.0 \text{ mm i.d.}$	<i>7</i> 32196	\$1,250.00
Chiral AGP	5 μm	15 cm x 2.0 mm i.d.	<i>7</i> 32197	\$1,400.00
Chiral AGP	5 μm	5 cm x 4.0 mm i.d.	<i>7</i> 32198	\$995.00
Chiral AGP	5 μm	$10 \text{ cm} \times 4.0 \text{ mm i.d.}$	<i>7</i> 32200	\$1,230.00
Chiral AGP	5 μm	15 cm x 4.0 mm i.d.	<i>7</i> 32199	\$1,490.00
Chiral AGP	5 μm	10 cm x 10.0 mm i.d.	<i>7</i> 32301	\$4,750.00
Chiral AGP	5 μm	15 cm x 10.0 mm i.d.	732302	\$ <i>7</i> ,100.00
Chiral AGP				
Guard Column	5 μm	1 cm x 3.0 mm i.d.	732300	\$135.00
Guard cartridge ho	older		<i>7</i> 31441	\$125.00

Omeprazole

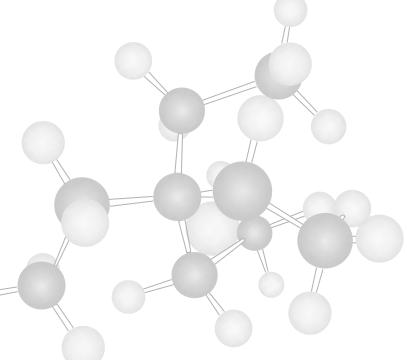
Column: Chiral-AGP

 $10 \text{ cm} \times 4.0 \text{ mm} \text{ i.d.}$

Mobile phase: 10% acetonitrile in 10 mM sodium

phosphate buffer, pH 6.5

Detection: UV 210 nm Sample Conc: 0.02 mg/mL Run Time: 8 min



PROTEIN-BASED CHIRAL STATIONARY PHASES

Chiral CBH

Micro, Analytical and Semi-Preparative Columns

> Cellobiohydrolase (CBH) is a stable enzyme which has been immobilized onto 5 µm spherical silica particles. The column is used in reversed-phase mode and is effective for the separation of enantiomers of basic drugs from many compound classes. The retention and the enantioselectivity can be regulated by changes in pH, buffer concentration and the nature and concentration of organic modifer.

Product	Particle Size	Column Length and i.d.	Product #	U.S. Price
Chiral CBH	5 μm	$10 \text{ cm} \times 0.18 \text{ mm} \text{ i.d.}$	732357	\$1,550.00
Chiral CBH	5 μm	10 cm x 2.0 mm i.d.	732353	\$1,250.00
Chiral CBH	5 μm	15 cm x 2.0 mm i.d.	732354	\$1,400.00
Chiral CBH	5 μm	$5 \text{ cm} \times 4.0 \text{ mm i.d.}$	732352	\$995.00
Chiral CBH	5 μm	$10 \text{ cm } \times 4.0 \text{ mm i.d.}$	<i>7</i> 32350	\$1,230.00
Chiral CBH	5 μm	15 cm x 4.0 mm i.d.	732351	\$1,490.00
Chiral CBH	5 μm	$10 \text{ cm} \times 10.0 \text{ mm i.d.}$	732355	\$4,750.00
Chiral CBH	5 μm	$15~\text{cm} \times 10.0~\text{mm}$ i.d.	732356	\$ <i>7</i> ,100.00
Chiral CBH				
Guard Column	5 μm	1 cm x 3.0 mm i.d.	732358	\$135.00
Guard cartridge ho	older		731441	\$125.00

Octopamine

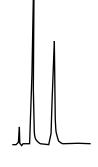
Column: Chiral-CBH

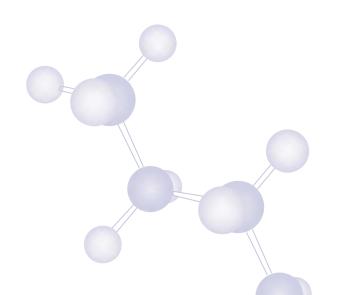
 $10 \text{ cm} \times 4.0 \text{ mm} \text{ i.d.}$

Mobile phase: 5% 2-propanol in 10 mM

sodium phosphate buffer, pH 6.0 + 50 µM disodium EDTA

Sample Conc: 0.03 mg/mL**Run Time:** $9 \ min$





PROTEIN-BASED CHIRAL STATIONARY PHASES

Chiral HSA

Analytical and Semi-Preparative Columns

With the Chiral human serum albumin (HSA) column, the enantiomers of many carboxylic acids and amino acids can be resolved directly, without derivatization. Enantioselectivity and retention can be regulated by changing the mobile phase composition, pH, buffer concentration and/or nature of the organic modifier.

HSA has been immobilized onto $5~\mu m$ spherical silica particles. The surface chemistry of the silica and the method of immobilization provide a stable chiral separation material.

Product	Particle Size	Column Length and i.d.	Product #	U.S. Price
Chiral HSA	5 μm	$10 \text{ cm} \times 4.0 \text{ mm} \text{ i.d.}$	732240	\$1,230.00
Chiral HSA	5 μm	15 cm x 4.0 mm i.d.	732239	\$1,490.00
Chiral HSA	5 μm	$10 \text{ cm} \times 10.0 \text{ mm} \text{ i.d.}$	732341	\$3,775.00
Chiral HSA	5 μm	15 cm x 10.0 mm i.d.	732342	\$5,050.00
Chiral HSA				
Guard Column	5 μm	1 cm x 3.0 mm i.d.	732340	\$135.00
Guard cartridge holder			731441	\$125.00

Mephenytoin

28

Column: Chiral-HSA

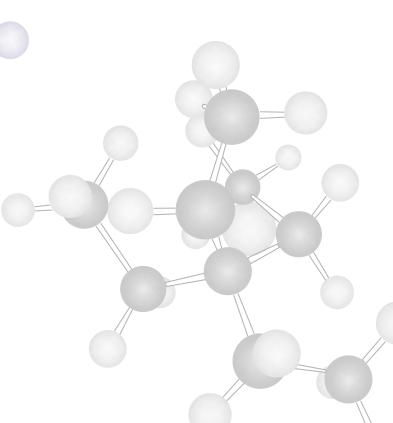
10 cm x 4.0 mm i.d.

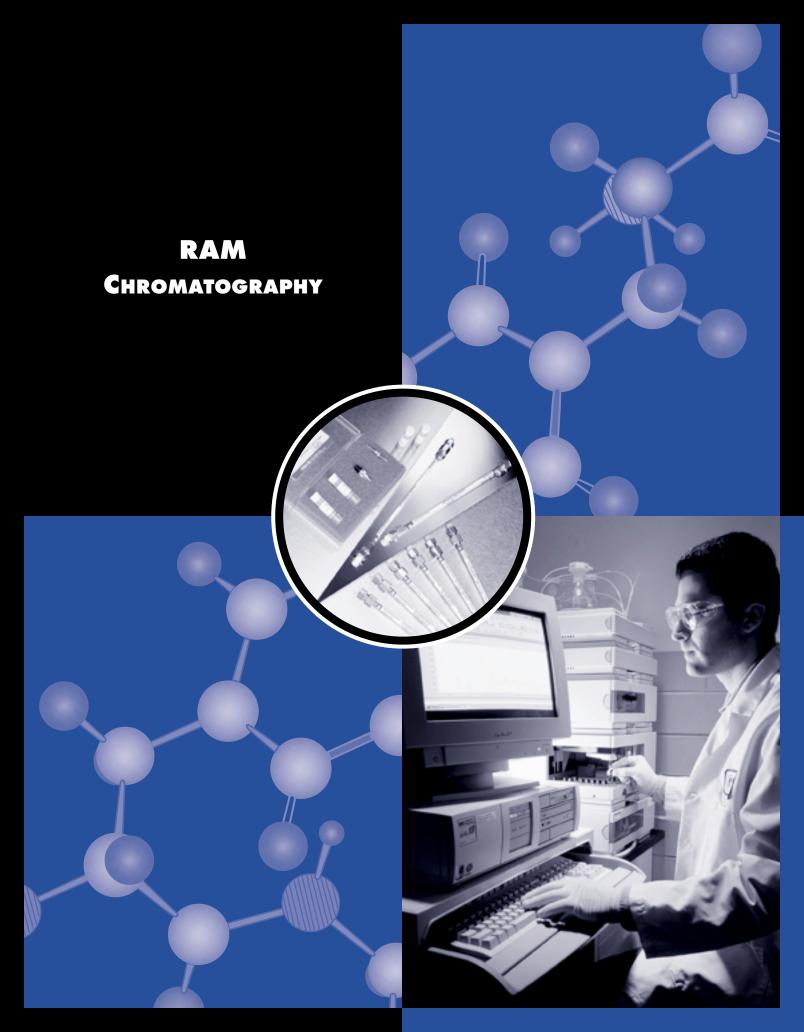
Mobile phase: 10 mM sodium phosphate buffer, pH 7.0

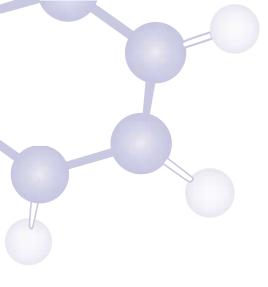
Detection: UV 225 nm Sample Conc: 0.02 mg/mL Run Time: 8 min



(800) 323-8144 ext. 649 (847) 967-6000 ext. 649 e-mail us at: sales@registech.com.







Conventional **RAM Concept** Sample Sample Precipitate proteins Centrifuge Solvent extract drugs from supernatant Evaporate extracting solvent Dissolve residue in mobile phase Filter sample Filter sample through through 0.2 μm filter 0.2 μm filter Inject onto Inject onto conventional RAM column **HPLC** column

Figure 1. RAM Direct Injection eliminates the lengthy pretreatment steps needed in conventional methods.

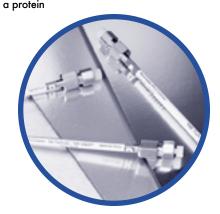
RAM DIRECT INJECTION (RESTRICTED ACCESS MEDIA)

A Tool for the Separation of Small Molecules in the Presence of Large Biomolecules

HPLC analysis of small molecules contained within a protein matrix can be a difficult and time consuming task.

The analysis often involves multi-step pretreatment procedures including centrifugation, extraction and filtration. RAM Direct Injection allows for the chromatographic resolution of small molecules in the presence of much larger analytes without extensive sample pretreatment (figure 1).

With RAM Direct Injection HPLC columns, a variety of complex sample matrices can be injected directly, without prior sample clean-up, and drugs, drug metabolites, peptides and other analytes can be separated and detected.

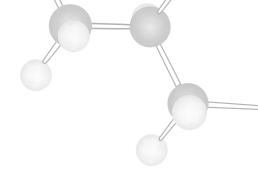


RAM Direct Injection Advantages

RAM Direct Injection technology offers the following benefits:

- Elimination of multiple sample pretreatment steps
 The use of RAM Direct Injection HPLC columns eliminates the precipitation, centrifugation, solvent evaporation and residue dissolution steps (Figure 1) of typical procedures. Simply
- dissolution steps (figure 1) of typical procedures. Simply filter the sample and inject directly onto the column.

 Useful with a variety of sample matrices
 - The RAM Direct Injection HPLC columns have demonstrated efficacy in the analysis of drugs, drug metabolites, peptides and other analytes in matrices such as plasma, serum, whole blood, urine, plant and tissue extract, food and beverage, and environmental samples.
- Compatibility with automated sample processing Simplified sample preparation and use of HPLC columns allows the use of automated systems.
- Reduction of potentially dangerous sample handling
 With direct injection, sample handling is significantly reduced;
 therefore, potentially dangerous samples such as plasma, serum,
 urine and environmental samples do not pose as significant
 a threat to the worker.



Reduction of biohazardous waste

Use of SPE (solid phase extraction) disks can create biohazardous waste. RAM Direct Injection columns limit the creation of unnecessary biohazardous waste.

• The lowest cost solution

Because of the benefits described above, RAM Direct Injection often offers the lowest total cost solution.

RAM Direct Injection Phases

Porous silica supports are characterized as consisting of an external, directly accessible surface and internal pores accessible only to molecules with an approximate molecular weight of less than 12,000 Daltons. Most conventional HPLC phases have a homogenous stationary phase on both silica surfaces. In contrast, the RAM phases are prepared by unique bonding processes that result in distinct inner and outer surfaces. A dual surface configuration is especially important because the majority of the silica's surface area is in the pores.

This dual phase system allows for the separation of analytes through a combination of size exclusion and conventional phase partitioning. The outer surface employs both size exclusion and hydrophilic interaction to prevent large biomolecules from accessing the inner layer. As a result, these compounds elute from the column at the void volume.

Small molecules penetrate through to the inner surface where they are retained and separated by the underlying hydrophobic support.

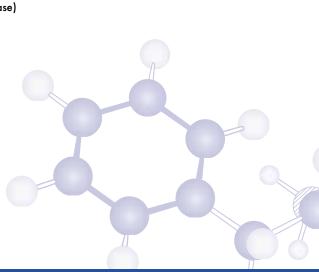
There are two RAM Direct Injection Technologies:

ISRP (Internal Surface Reversed Phase)

- GFF
- GFF II

SPS (Semi-Permeable Surface)

- Nitrile
- Octyl (C8)
- ODS (C18)
- Phenyl



ISRP (INTERNAL SURFACE REVERSED PHASE)

Developed by Dr. Thomas Pinkerton, this material was created specifically for the direct analysis of drugs in serum without extensive sample preparation. The result was a new phase that allows for chromatographic separations without interference by protein adsorption.

GFF

The GFF name is derived from the glycine-L-phenylalanine-L-phenylalanine tripeptide bonded within the silica pores. GFF was chosen from many available peptides for its selectivity towards positively charged aromatic analytes and a variety of neutral molecules. The hydrophilic outer surface, created from the cleavage of the original tripeptide, is comprised of the single amino acid, glycine.

GFF II

Continuing product improvement efforts resulted in the development of the ISRP GFF II, a second generation phase with an improved bonding process—bonding the GFF peptide to the silica surface through a monofunctional glycidoxypropyl linkage rather than the original trifunctional linkage. This resulted in the following improvements:

- Increased sample retention
- Higher column efficiency
- Greater batch-to-batch reproducibility

ISRP Selectivity

Many variables can affect the selectivity of the ISRP phase, including:

Mobile Phase Composition:

The nature of ISRP analytes requires that mobile phases consist of a buffer with varying degrees of modification. Modifiers can include acetonitrile, methanol, isopropanol, and tetrahydrofuran. Caution: too much modifier can result in matrix precipitation.

pH: The pH of the mobile phase can be controlled to avoid protein denaturing and to enhance selectivity. The pH range of the column is between 2.5 and 7.5; however, within the optimal pH range of 6.0 to 7.5, both the proteins and the glycine outer surface take on a negative charge. As a result, negatively charged proteins are repulsed by the outer phase, and pass quickly through the column.

Temperature: Separations can also be optimized by varying column temperature. Lower temperatures have been shown to result in increased retention and selectivity.

Product	Particle Size	Column Length and i.d.	Product #	U.S. Price
ISRP GFF and GFF II Analytical Columns				
GFF	5 μm, 80Å	5 cm x 2.1 mm i.d.	731449	\$495.00
GFF	5 μm, 80Å	5 cm x 4.6 mm i.d.	731450	\$535.00
GFF	5 μm, 80Å	$15 \text{ cm} \times 4.6 \text{ mm} \text{ i.d.}$	<i>7</i> 31451	\$ <i>7</i> 95.00
GFF	5 μm, 80Å	25 cm x 4.6 mm i.d.	731452	\$1,045.00
GFF II	5 μm, 80Å	5 cm x 2.1 mm i.d.	731469	\$495.00
GFF II	5 μm, 80Å	5 cm x 4.6 mm i.d.	731470	\$535.00
GFF II	5 μm, 80Å	15 cm x 4.6 mm i.d.	731471	\$795.00
GFF II	5 μm, 80Å	25 cm x 4.6 mm i.d.	731472	\$1,045.00

Particle Size	Column Length and i.d.	Product #	U.S. Price	
ISRP GFF and GFF II Guard Columns				
5 μm, 80Å	1 cm x 3.0 mm i.d.	731440	\$260.00	
acements* *		<i>7</i> 31444	\$175.00	
5 μm, 80Å	1 cm x 3.0 mm i.d.	<i>7</i> 31475	\$260.00	
placements**		731474	\$1 <i>7</i> 5.00	
Guard cartridge holder		731441	\$125.00	
		<i>7</i> 31443	\$35.00	
	F II Guard Colum 5 μm, 80Å acements** 5 μm, 80Å placements**	F II Guard Columns 5 μm, 80Å 1 cm x 3.0 mm i.d. accements** 5 μm, 80Å 1 cm x 3.0 mm i.d. placements**	F II Guard Columns 5 μm, 80Å 1 cm x 3.0 mm i.d. 731440 accements** 731444 5 μm, 80Å 1 cm x 3.0 mm i.d. 731475 placements** 731474 Ider 731441	

^{*}Includes 1 holder, 1 coupler, 2 guard cartridges

^{**}Includes 3 guard cartridges

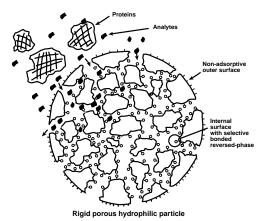
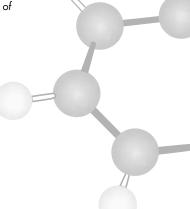


Figure 2. Demonstrates the inner and outer layers of a typical ISRP phase.



SPS (SEMI-PERMEABLE SURFACE)

In an effort to extend the applicability of the RAM Direct Injection columns, Regis, in conjunction with Dr. Fred Regnier and Dr. Carla Desilets at Purdue University, developed the Semi-Permeable Surface (SPS) phases.

SPS Structure

Like the ISRP phase, the SPS phases consist of both hydrophilic outer and hydrophobic inner surfaces. The distinct difference is that the inner and outer surfaces of the SPS are bonded separately, allowing each to be varied independently. The SPS structure includes a hydrophobic inner phase such as ODS, and a hydrophilic outer phase of polyethylene glycol (figure 3). The outer phase provides size exclusion and hydrophilic shielding, which repels large biomolecules. The various inner phases allow for separation of small analytes.

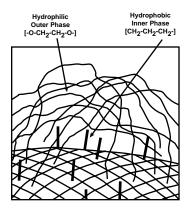


Figure 3. SPS structure highlighting the inner and outer phases.

Product	Particle Size	Column Length and i.d.	Product #	U.S. Price
SPS Analytical Co				
Nitrile	5 μm, 100Å	15 cm x 4.6 mm i.d.	785105	\$500.00
Nitrile	5 μm, 100Å	$25 \text{ cm} \times 4.6 \text{ mm} \text{ i.d.}$	785205	\$750.00
Phenyl	5 μm, 100Å	$15 \text{ cm} \times 4.6 \text{ mm i.d.}$	785107	\$500.00
Phenyl	5 μm, 100Å	$25 \text{ cm} \times 4.6 \text{ mm i.d.}$	785207	\$750.00
Octyl	5 μm, 100Å	5 cm x 2.1 mm i.d.	785308	\$330.00
Octyl	5 μm, 100Å	$5 \text{ cm} \times 4.6 \text{ mm i.d.}$	785008	\$385.00
Octyl	5 μm, 100Å	$15 \text{ cm} \times 4.6 \text{ mm i.d.}$	785108	\$500.00
Octyl	5 μm, 100Å	$25 \text{ cm} \times 4.6 \text{ mm} \text{ i.d.}$	785208	\$750.00
ODS	5 μm, 100Å	5 cm x 2.1 mm i.d.	785318	\$330.00
ODS	5 μm, 100Å	5 cm x 4.6 mm i.d.	785018	\$385.00
ODS	5 μm, 100Å	15 cm x 4.6 mm i.d.	<i>7</i> 85118	\$500.00
ODS	5 μm, 100Å	25 cm x 4.6 mm i.d.	785218	\$750.00

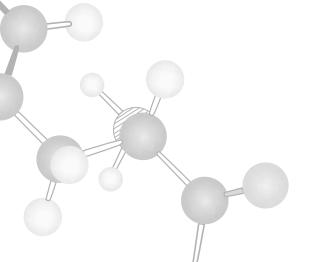
Product	Particle Size	Column Length and i.d.	Product #	U.S. Price
SPS Guard Columns				
Nitrile Guard Kit*	5 μm, 100Å	1 cm x 3.0 mm i.d.	785405	\$250.00
Nitrile Guard repla	cements**		785505	\$170.00
Phenyl Guard Kit*	5 μm, 100Å	$1 \text{ cm} \times 3.0 \text{ mm i.d.}$	785407	\$250.00
Phenyl Guard replacements**			785507	\$170.00
Octyl Guard Kit*	5 μm, 100Å	$1~\mathrm{cm} \times 3.0~\mathrm{mm}$ i.d.	785408	\$250.00
Octyl Guard replac	cements**		785508	\$170.00
ODS Guard Kit*	5 μm, 100Å	$1 \text{ cm} \times 3.0 \text{ mm i.d.}$	785418	\$250.00
ODS Guard replac	ements**		785518	\$1 <i>7</i> 0.00
Guard cartridge ho	older		<i>7</i> 31441	\$125.00
Coupling			<i>7</i> 31443	\$35.00

^{*}Includes 1 cartridge holder, 1 coupling, and 2 guard cartridges

^{**}Includes 3 guard cartridges

Particle Size	Column Length and i.d.	Product #	U.S. Price
lge Columns			
Kit* 5 μm, 100Å	1 cm x 10.0 mm i.d.	785608	\$400.00
replacements * *		785708	\$380.00
(it* 5 μm, 100Å	1 cm x 10.0 mm i.d.	785618	\$400.00
eplacements**		785718	\$380.00
holder		731446	\$200.00
	Particle Size dge Columns Kit* 5 μm, 100Å replacements** Kit* 5 μm, 100Å eplacements** holder	dge Columns Kit* 5 μm, 100Å 1 cm x 10.0 mm i.d. replacements** Xit* 5 μm, 100Å 1 cm x 10.0 mm i.d. eplacements** 1 cm x 10.0 mm i.d.	dge Columns Kit* 5 μm, 100Å 1 cm x 10.0 mm i.d. 785608 replacements** 785708 Kit* 5 μm, 100Å 1 cm x 10.0 mm i.d. 785618 eplacements** 785718

^{*}Includes 1 cartridge holder, 1 coupling, and 2 guard cartridges



^{**}Includes 3 guard cartridges

SPS (SEMI-PERMEABLE SURFACE)

SPS Column Advantages

The SPS offers the following advantages:

- Increased durability
- Increased selectivity
- Allows use of buffered, normal-phase, and reversed-phase systems

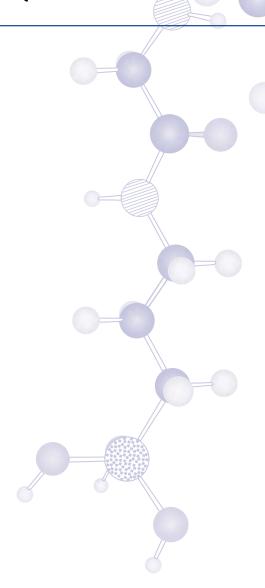
SPS Selectivity

The primary advantage of SPS over ISRP GFF is that the inner surface of SPS may be varied independently of the outer, resulting in a wider scope of analysis opportunities. Available inner phases include the following:

- Nitrile
- Octyl (C8)
- ODS (C18)
- Phenyl

The retention mechanism of these SPS phases involves hydrogen bonding by the outer phase and hydrophobic interaction by the inner phase. Polar solutes interact primarily with the outer phase and show little discrimination among the various inner phases. Conversely, the nonpolar solutes interact primarily with the inner phase.

The SPS phases allow use of buffered, normalphase, and reversed-phase eluents. The actual composition is limited only by the pH and organic modifier parameters dictated by the proteins contained within the sample.



RAM DIRECT INJECTION APPLICATIONS

RAM Direct Injection has been effective in numerous applications. Adjacent is a listing of some of the compounds analyzed by RAM Direct Injection. For additional applications, please contact Regis for the RAM Direct Injection Application Guide or download from Regis Web site at www.registech.com/ram/.

Some Compounds Analyzed by RAM Methods						
Acetazolamide	Cefaclor	Oxyphenbutazone	Sulfinpyrazone			
Acetaminophen	Cefpiramide	Pentobarbital	Tamoxifen			
Acetylsalicylic acid	3,4-Diaminopyridine	Phenelzine	Theophylline			
4-Aminopyridine	Furosemide	Phenobarbital	Trazodone			
Amobarbital	Heparin	Phenylalanine	Trimethoprim			
Aprotinin	Hydroxyzine	Phenylbutazone	Trimipramine			
Barbital	Imipramine	Phenytoin	Tryptophan			
Butabarbital	Imirestat	Propranolol	Tyrosine			
Caffeine	Methyl salicylate	Salicylic Acid	Verapamil			
Carbamazepine	Norverapamil	Secobarbital	Warfarin			

RAM DIRECT INJECTION APPLICATION/ISRP

Separation of Barbiturates in Human Serum

Column: ISRP GFF II, 5 μm, 80Å

15 cm x 4.6 mm i.d.

Mobile Phase: (95/5) 0.1 M potassium

phosphate buffer, pH 7.5/

methanol

 $\begin{tabular}{llll} Flow Rate: & 1.0 mL/min \\ Load: & 10 \ \mu L \\ \hline \end{tabular}$

Sample Composition

in Human Serum:

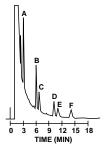
A. Barbital

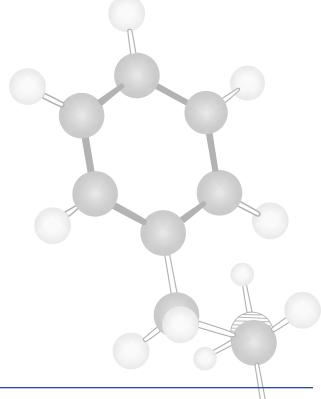
B. Phenobarbital

C. ButabarbitalD. Amobarbital

E. Pentobarbital

F. Secobarbital





RAM DIRECT INJECTION APPLICATION/SPS

Determination of Antipyrine and Acetaminophen

Column: SPS C8, 5 μm, 100Å

25 cm x 4.6 mm i.d.

Mobile Phase: (99/0.5/0.5) 0.1 M potassium

phosphate buffer, pH 7.4/ acetonitrile/tetrahydrofuran

Flow Rate: 1.0 mL/min, 37° C

Load: $25 \mu l$ Detection: UV 244 nm

Sample Composition

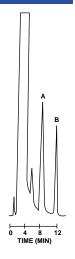
in Human Serum: A. Antipyrine

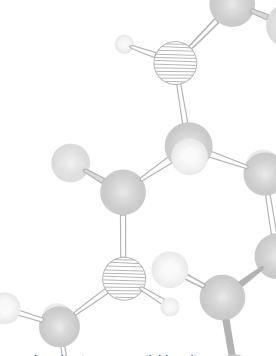
B. Acetaminophen

Reference: Gurley, B.J.: et al.; Determination

of Antipyrine in Human Serum by Direct Injection Restricted Access Media Liquid Chromatography; J. Pharm. Biomed Anal. 1994, 12 (12) 1591–1595

12 (12), 1591 - 1595.





RAM DIRECT INJECTION APPLICATIONS/COLUMN SWITCHING

Column Switching with RAM Columns

Column Switching for Improved Sensitivity

There has been growth in the use of column switching to process a large number of samples and achieve high sensitivity. The RAM Direct Injection column can be used in a column switching application to retain the small nonpolar analytes while allowing the matrix to pass through to waste. A less polar organic mobile phase is then used to elute the accumulated analytes onto an analytical column for subsequent chromatography.

Recent column switching work involves the use of short RAM guard columns. The guard column is used to separate the analytes from the matrix before switching to an analytical column. The low cost of the guard column allows it to be discarded after 60 to 100 samples. The RAM guard column is an inexpensive and simpler alternative to Solid Phase Extraction.

Figure 4 depicts a typical column switching system. In this procedure, the prefiltered but otherwise untreated sample is injected directly onto a RAM column. In the RAM column the smaller molecules are retained and concentrated, while most of the larger molecules are passed to waste. A stronger mobile phase is then used to elute the analytes onto a second column – often octadecylsilyl (ODS) – where they are separated and analyzed.

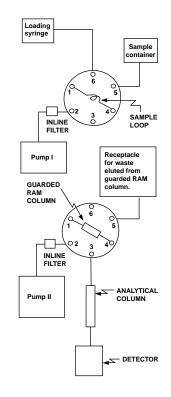
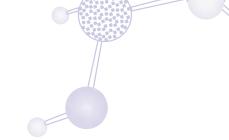
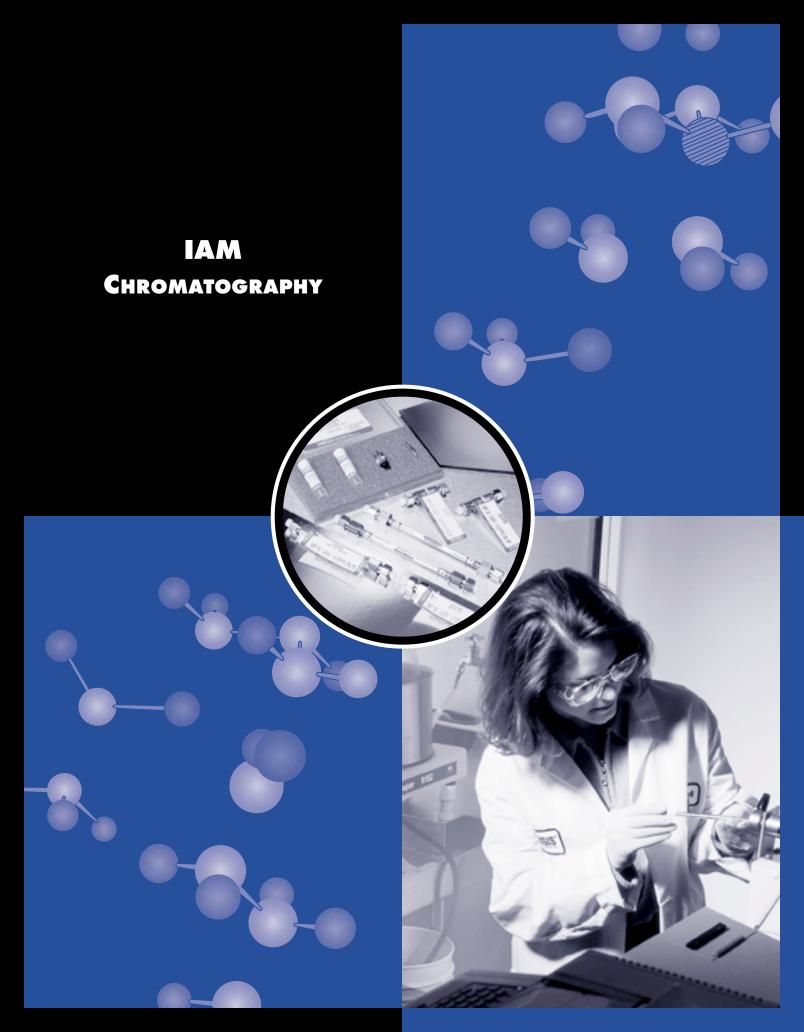


Figure 4. Column switching system.

For additional information on RAM Direct Injection, check our Web site at www.registech.com/ram/. You may also request a copy of the RAM Direct Injection Application Guide containing additional RAM applications, by contacting Regis at:

(800) 323-8144 ext. 649 (847) 967-6000 ext. 649 e-mail us at: sales@registech.com.





IAM CHROMATOGRAPHY

Immobilized Artificial
Membrane (IAM)
technology is an
innovative approach
to chromatography in
which the chromatographic
surface emulates the
lipid environment of
the cell membrane. 1,2

HPLC Separation Tools for Membrane Protein Purification and Drug Membrane Permeability Prediction

Phosphatidylcholine (PC) is the major phospholipid found in cell membranes. IAM chromatography phases prepared from PC analogs closely mimic the surface of a biological cell membrane. Consequently, IAM phases display a high affinity for membrane proteins and are useful in membrane protein purification and in the study of drug-membrane interactions. The IAM surface is formed by covalently bonding the membrane-forming phospholipids to silica.



Several different types of IAM columns are used for various applications:

Membrane Protein Purification

IAM.PC

IAM.PC.MG

Drug Discovery

IAM.PC.DD2

- Drug membrane permeability prediction
- Hydrophobic in nature

IAM Fast-Screen Mini Columns

High throughput estimation of drug permeability



MEMBRANE PROTEIN PURIFICATION

IAM.PC

The IAM.PC phase, developed by Dr. Charles Pidgeon of Purdue University, was the first in a line of IAM phases to be manufactured by Regis. Use of this phase has simplified the inherent difficulties of protein isolation and purification, ³⁹ allowing for rapid purification of membrane proteins while maintaining biological activity. The IAM.PC phase is an important tool for the pharmaceutical industry and academia alike.

The first IAM stationary phase was based on the prevalent membrane lipid, phosphatidylcholine (PC), and consists of monolayers of amphiphilic phospholipids covalently bonded to aminopropyl silica particles through a terminal amide linkage. As a result, the bulky phosphatidylcholine groups shield many of the amine binding sites on the silica surface, preventing amine interaction with the protein molecules.

The membrane nature of the IAM phase imparts surface characteristics which are useful in the chromatography of membrane proteins. These include: high protein loading, increased protein recovery, recovery of functional activity, and selectivity for membrane proteins.

Large membrane proteins can interact with any combination of polar headgroup, hydrophobic chain, or inner amine groups. The subsurface has been shown to interact with certain solutes, and may or may not contribute to the separation of a given biomolecule. The residual amines can be left unaltered on the subsurface or deactivated through an endcapping procedure, which results in increased stability of the bonded phase.⁷ The methyl glycolate endcapping, for example, converts residual amines to neutral amides and introduces a hydroxyl group (IAM.PC.MG).

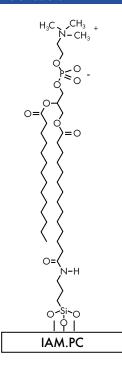
IAM.PC Applications

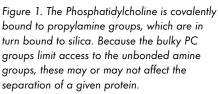
Numerous applications have been developed using IAM.PC columns:

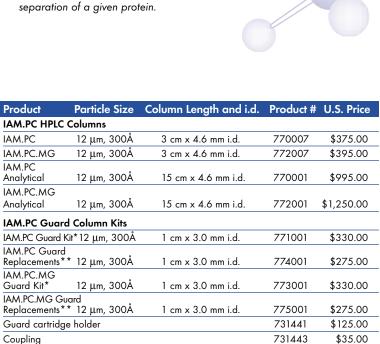
- Purification of Cytochrome P450
- Isolation of membrane proteins
- Prediction of solute transport across human skin
- Prediction of amino acid transport across the blood-brain barrier
- Binding of solutes to liposome membranes
- Immobilization of Trypsin and α-chymotrypsin for the determination of their inhibitor and substrate activity

For additional information on IAM.PC applications please contact Regis' technical support staff.

IAM PC Structure







^{*} Includes 1 cartridge holder, 1 coupling, 2 guard cartridges

^{**} Includes 3 guard cartridges

DRUG DISCOVERY - PREDICTING DRUG MEMBRANE PERMEABILITY

IAM.PC.DD2 IAM Fast - Screen Mini Column

IAM chromatography has recently gained acceptance among drug discovery chemists for estimating the membrane permeability of small molecule drugs.

Figure 2 illustrates that the interaction between membrane bilayer and drug can be modeled by the IAM column/drug system.

 K_{IAM} , the equilibrium constant describing the relative concentrations of drug in the membrane and in the external fluid, is analogous to the k'_{IAM} .

This IAM technique provides superior correlation with experimentally determined drug permeability when compared to other chromatographic methods. ODS silica, for example, retains analytes solely on the basis of hydrophobicity. IAM more closely mimics the interaction of analytes with biological membranes, where a combination of hydrophobic, ion pairing, and hydrogen bonding interactions are possible. This combination of interactions measured by the IAM column is known as phospholipophilicity.

These advances have led to the development of several new IAM phases used for predicting drug membrane permeability:

- IAM.PC.DD2
- IAM Fast-Screen Mini Column

Intestinal Drug Permeability

The retention factors measured on reversed phase C18 (ODS) columns (a commonly used model to determine drug partitioning) show extremely poor correlation with intestinal drug absorption (figure 3). For this group of compounds, hydrophobicity alone, as measured by the reversed-phase C18 column, is a poor predictor of drug absorption. Since IAM.PC Drug Discovery columns measure both hydrophilic and hydrophobic interactions between drugs and membranes, the IAM.PC Drug Discovery Column is better suited to the prediction of intestinal drug absorption.

Fluid Membrane and IAM Drug Partitioning Measurements

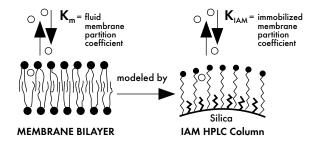


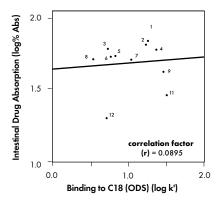
Figure 2. Fluid membrane bilayer can be modeled by IAM column.

ODS Exhibits Poor Correlation with Intestinal Drug Absorption

Column: C18 (ODS) 3 cm x 4.6 mm i.d.

Mobile Phase: 0.01 M DPBS Buffer, pH 5.4

Flow Rate: 1.0 mL/min Load: 10 μ L Detection: UV 220 nm



- 5. Aniline6. *m*-Nitrobenzoic acid
- 7. Phenol
- Benzoic Acid
 Acetanilide

1. m-Nitroaniline

2. p-Nitroaniline

3. Salicylic acid

4. p-Toluidine

- 10. Antipyrine
- 11. Theophylline12. Acetylsalicylic acid

Figure 3. Drug partitioning into ODS does not correlate with intestinal drug absorption.

IAM.PC.DD2

Like the first generation IAM.PC.DD material, the IAM.PC.DD2 is used to predict drug membrane permeability. The ester bonding of the DD2 packing offers more hydrophobicity than the first generation DD phase. This material is a diacylated or double chain ester PC ligand and is endcapped with C10/C3 alkyl chains as illustrated in figure 4.

Column Advantages

The IAM.PC.DD2 material offers the following advantages:

- Hydrophobic nature
- Greater stability
- Excellent correlation to traditional methods

Hydrophobic Nature

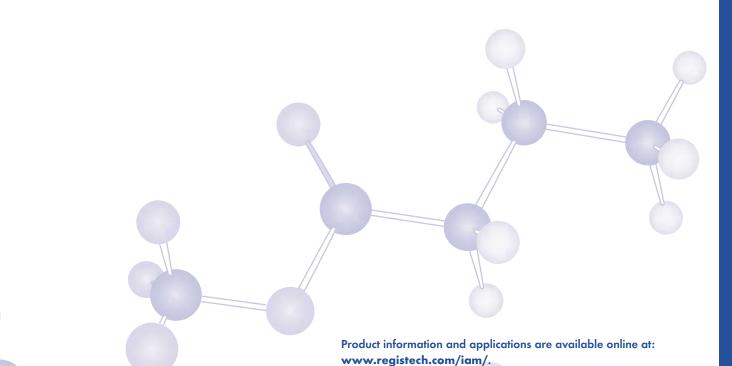
The IAM.PC.DD2 offers more hydrophobicity than the first generation IAM.PC.DD material. This hydrophobic nature allows for longer retention times to compounds not well retained on the IAM.PC.DD material.

Greater Stability

Another distinct advantage of the IAM.PC.DD2 material is its ability to tolerate mobile phases between pH's 7.0 to 7.5, thus resulting in longer column life under these conditions.

IAM.PC.DD2 Structure

Figure 4. IAM.PC.DD2 is used to predict drug membrane permeability.



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IAM.PC.DD2

Excellent Correlation to Traditional Methods

The traditional means of predicting membrane permeability include the use of Caco-2 cell line cultures, intestinal tissue or liposome assays. These methods are laborious and costly to perform.

Intestinal Tissue Correlation

Measuring drug permeability in the intestinal tissue, where absorption is occurring, is physiologically more relevant than measuring drug permeability in Caco-2 cells. Figure 5 and table 1 illustrate that drug absorption in this inverted rat intestinal tissue model correlates with drug retention factors k'_{IAM} measured on the IAM.PC.DD2 column.

Sample	% Absorption of Inverted Rat Intestine	(k′) IAM.PC.DD2
m-nitroaniline	77	10.838
p-nitroaniline	68	16.086
salicylic acid	60	6.963
p-toluidine	59	4.546
aniline	54	2.069
m-nitrobenzoic acid	53	4.403
phenol	51	6.544
benzoic acid	51	2.088
acetanilide	42	5.096
antipyrine	32	3.350
theophylline	29	1.478
acetylsalcylic acid	20	0.931
r (correlation factor)*		0.8025

^{*} r is calculated by plotting log k' vs. log % absorption of inverted rat intestine.

Table 1. Correlating Drug Partitioning into IAM with rat intestinal drug absorption.

Product	Particle Size	Column Length and i.d.	Product #	U.S. Price
IAM.PC.DD2	Columns			
IAM.PC.DD2	12 μm, 300Å	3 cm x 4.6 mm i.d.	<i>77</i> 4010	\$550.00
IAM.PC.DD2	12 μm, 300Å	10 cm x 4.6 mm i.d.	<i>77</i> 4011	\$975.00
IAM.PC.DD2	12 μm, 300Å	$15 \text{ cm} \times 4.6 \text{ mm} \text{ i.d.}$	<i>77</i> 4014	\$1,350.00
IAM.PC.DD2	12 μm, 300Å	Guard Kit*	<i>77</i> 4012	\$330.00
IAM.PC.DD2	12 μm, 300Å	Guard Replacements**	<i>77</i> 4013	\$275.00

^{*} Includes 1 holder, 1 coupler, 2 guard cartridges

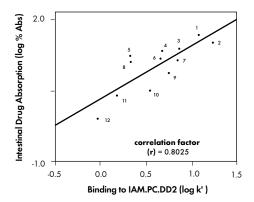
IAM Correlates with Intestinal Drug Absorption

Column:	AM.PC.DD2
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 $10~\text{cm} \times 4.6~\text{mm} \text{ i.d.}$

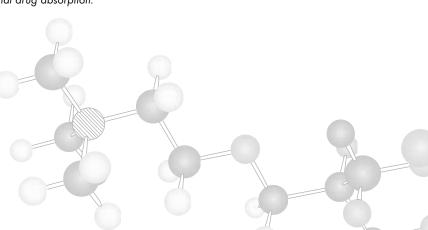
Mobile Phase: 0.01 M DPBS Buffer, pH 5.4

Flow Rate: 1.0 mL/min Load: 10 μ L Detection: UV 220 nm



- 1. m-Nitroaniline
- 2. p-Nitroaniline
- 3. Salicylic acid
- 4. p-Toluidine
- 5. Aniline
- 6. m-Nitrobenzoic acid
- 7. Phenol
- 8. Benzoic acid
- 9. Acetanilide
- 10. Antipyrine
- 11. Theophylline
- 12. Acetylsalicylic acid

Figure 5. IAM.PC.DD2 columns measure drug absorption in inverted rat intestinal tissue.



^{**} Includes 3 guard cartridges

Packed with the Ester PC Ligand phase, IAM Fast-Screen Mini columns are a rapid and economically viable screening method for the high throughput estimation of drug permeability. Their benefits include excellent reproducibility, short analysis time and low cost. This can be of great use in characterizing large libraries of compounds.

The structure of the esterIAM.PC.C10/C3 packing, selected for the Fast-Screen Mini Column, is shown in figure 6. This PC analog demonstrates superiority in retention times and stability — essential features for short columns and mass drug screening.

The IAM.PC Fast-Screen Mini Column, 1 cm in length by 3.0 mm in internal diameter, was specifically designed by Regis for rapid estimation of drug permeability in high throughput screening programs. When connected to an HPLC system with an autosampler, a single column can be used in the analysis of hundreds of samples per day with highly reproducible results.

The 1cm Fast-Screen Mini Column is offered not as a separation tool, but rather as a tool for characterizing the chromatographic retention factor (k') of individual analytes. The measured k' of analytes on this column can be used to estimate a value for drug permeability.

Column Advantages

Regis Technologies' 1 cm Fast-Screen Mini Column for Drug Discovery provides the following advantages:

- Excellent correlation to traditional methods
- Rapid indication of drug absorption
- High sample throughput
- Highly reproducible results
- Durability
- Cost effectiveness
- Ability to establish absorption zones for high throughput screening

IAM Fast-Screen Mini Column Structure

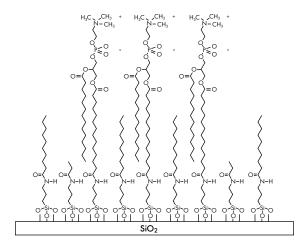


Figure 6. IAM.PC Fast-Screen Mini Column provides rapid estimation of drug permeability in high thoughput screening programs.



Excellent Correlation To Traditional Methods

The traditional means of predicting permeability include use of Caco-2 cell line cultures, intestinal tissue, or liposome assays. These are laborious and costly to perform.

Data obtained from the IAM Fast-Screen Mini Column correlate well to data obtained from traditional assays. This is summarized in table 2.

Method	Number of Compounds Evaluated	Correlation (r) with IAM Fast-Screen Mini Column
Partitioning into liposomes	23	0.831
Intestinal drug permeability	12	0.839
Caco-2 cell permeability	8	0.909

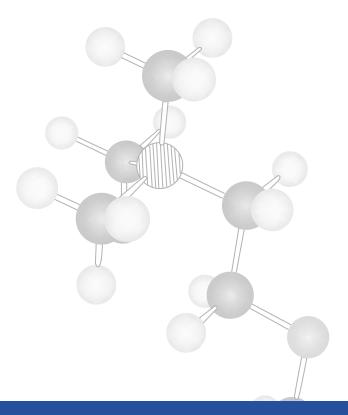
Table 2. Comparing k'_{IAM} data with other methods for estimating permeability.

Caco-2 Cell Correlation

Figure 7 illustrates that drug permeability predicted by Caco-2 cells correlates well to k'_{IAM} measured on the IAM Fast-Screen Mini Columns.

Intestinal Tissue Correlation

Table 3 shows that drug permeability predicted by Inverted Rat Intestines correlates well to drug retention factors, k'_{IAM} measured on the IAM Fast-Screen Mini Columns. Note the short retention times.

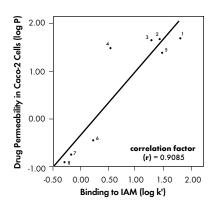


IAM Fast-Screen Correlates with Drug Permeability in Caco-2 Cells

Column: IAM Fast-Screen Mini Column 1 cm x 3.0 mm i.d

Mobile Phase: 0.01 M DPBS Buffer, pH 7.4

Flow Rate: 0.5 mL/min Load: 10 μ l Detection: UV 220 nm



- 1. Propranolol
- 2. Alprenolol
- 3. Warfarin
- 4. Metoprolol
- 5. Hydrocortisone
- 6. Terbutaline
- 7. Atenolol
- 8. (AVP) Arginine-Vasopressin

Figure 7. Correlating drug partitioning into IAM with intestinal drug permeability (log P) through Caco-2 cells.

	% Absorption of Inverted	IAM Fast-Screen Mini Column Retention	k′
Compound	Rat Intestine	Time (Sec)	(corrected)
m-nitroaniline	77	133.1	15.29
p-nitroaniline	68	1 <i>77</i> .9	21.84
salicylic acid	60	93.8	9.54
p-toluidine	59	79.7	7.48
aniline	54	52.1	3.45
m-nitrobenzoic acid	53	68.1	5.79
phenol	51	94.6	9.66
benzoic acid	51	43.7	2.22
acetanilide	42	<i>7</i> 6.2	6.97
antipyrine	32	51.8	3.40
theophylline	29	39.3	1.58
acetylsalcylic acid	20	36.1	1.11
r (correlation factor)*	= 0.8385		

^{*} r is calculated by plotting log k' vs. log % absorption of inverted rat intestine.

Chromatographic Conditions:

Column: IAM Fast-Screen Mini Column

1 cm x 3.0 mm i.d.

Mobile Phase: Dulbecco's Phosphate Buffered Saline, pH 5.4

Flow Rate: 0.3 mL/min

Detection: UV 254 nm, 0.1 AUFS

Table 3. Correlating drug partitioning into IAM with rat intestinal drug absorption.

Rapid Indication of Drug Absorption

IAM Chromatography is a more rapid alternative to other methods. In a recent study completed by Regis, k'_{IAMs} of 12 compounds were compared with absorption data obtained in situ using rat intestines. Retention times for the compounds tested were between 20 and 180 seconds, while retention factors correlated well to the intestinal absorption data.

High Sample Throughput

IAM chromatography is of increasing importance in combinatorial chemistry, where it is used to provide an initial estimate of a drug candidates' membrane permeability. Hundreds of samples can be injected into a single Fast-Screen Mini Column using an automated HPLC system. Recently a group of 12 test analytes was evaluated in 10 runs over the course of eight hours. Total run time for the 12 test analytes was only 42 minutes.

Highly Reproducible Results

The measured values for k'_{IAM} show excellent reproducibility, both from run to run and from day to day (figure 8).

Durability

IAM Fast-Screen Mini Columns are extremely durable. Correlation factors, r, for the original k', and k' after 5000 column volumes were identical.

Cost Effectiveness

Because the IAM Fast-Screen Mini Column is inexpensive, has a very short analysis time, and provides drug permeability estimates for hundreds of drug candidates in a fraction of the time of conventional methods, the IAM Fast-Screen Mini Column becomes the economical alternative for high throughput screening.

Ability to Establish Permeability Zones for High Throughput Screening

Permeability zones can be determined for different analytes when performing large-scale drug absorption screening. Thus, rapid IAM analyses can characterize a drug as having low, medium, or high membrane permeability (figure 9).

Excellent Reproducibility with IAM Fast-Screen Mini Column

Figure 8. Highly reproducible k'_{IAM} from 10 runs over a two-day period.

compound

k'IAM Permeability Zones

Permeability Zones

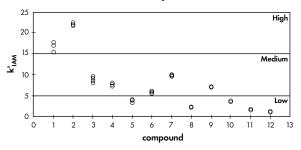
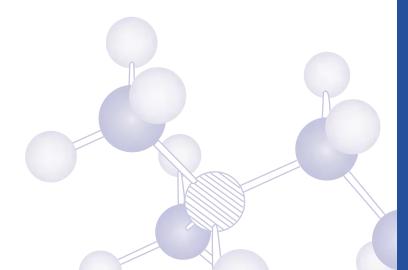


Figure 9. Permeability zones established large-scale drug absorption screening.



Regis Technologies manufactures the IAM Fast-Screen Mini Column on-site in its manufacturing facility. This column, as well as all of our other products, must adhere to rigorous manufacturing and quality control specifications before release.

Regis' technical support staff, with years of chromatography experience, is available to answer any questions regarding the new IAM Fast-Screen Mini Column.

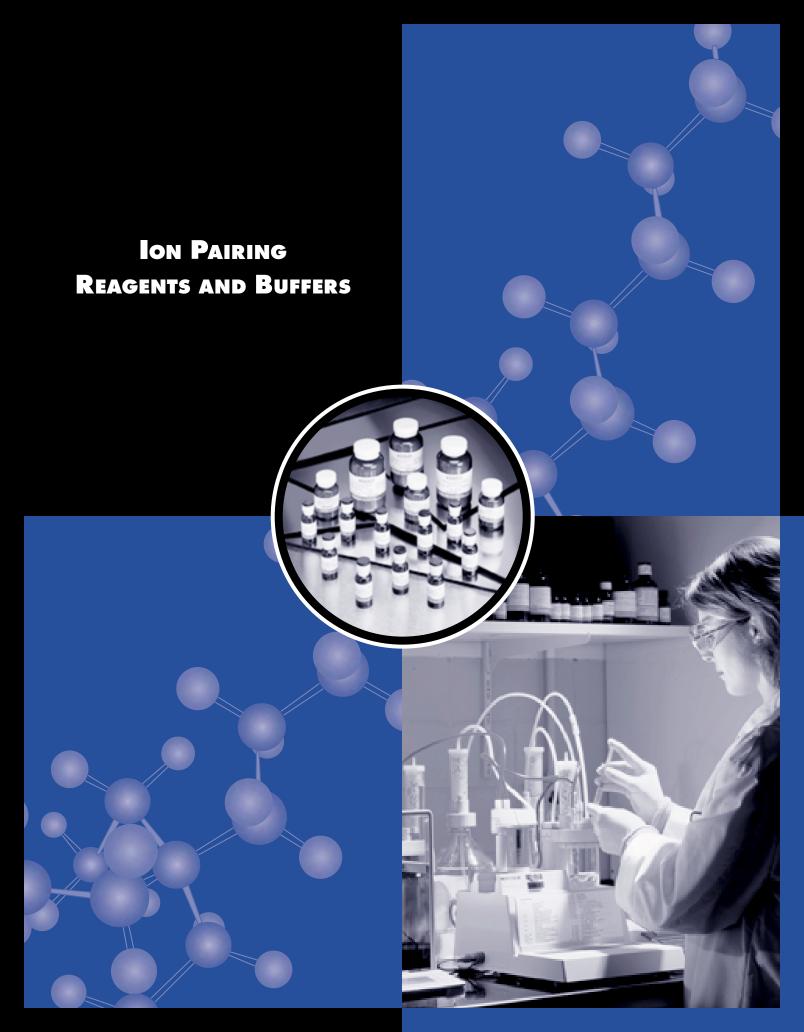
Product	Particle Size	Column Length and i.d.	Product #	U.S. Price
IAM Fast-Screen Mini Column Kit	12 μm, 300Å	1 cm x 3 mm i.d.	775014	\$375.00
 2 columns 1 cm x 3 mm 1 cartridge holder Two 1/16" nuts and ferrules Two end plugs Care and use booklet 				
IAM Fast-Screen Mini Columns, Pkg of 6 6 columns, 1 cm x 3 mm Care and use booklet	12 μm, 300Å	1 cm x 3 mm i.d.	<i>775</i> 01 <i>5</i>	\$575.00
IAM Fast-Screen Mini Columns, Pkg of 12 • 12 columns, 1 cm x 3 mm • Care and use booklet	12 μm, 300Å	1 cm x 3 mm i.d.	<i>77</i> 5016	\$1,000.00

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- 11. Schanker, L. S.; J. Pharmacol. Exp. Ther. 1958, 123, 81-88.

For additional information on IAM Chromatography, check our Web site at www.registech.com/iam/. You may also request a complete IAM publication list, by contacting Regis at:

(800) 323-8144 ext. 649 (847) 967-6000 ext. 649 e-mail us at: sales@registech.com.



ULTRAPURE ION PAIRING REAGENTS AND BUFFERS

Ion Pair Chromatography
is a method for improving
the separation of charged
analytes. In the resolution
of organic ions with
conventional HPLC methods,
use of ion pair reagents can
enhance peak shape and
retention time when common
remedies such as modifying
eluent ratios or changing
stationary phase fail.

The Advantages of Ion Pair Chromatography

In the past, chromatographic separation of charged analytes has been achieved by ion suppression (the careful adjustment of the mobile phase pH to result in a nonionized analyte). Determining the optimum mobile phase pH in ion suppression, however, often requires extensive method development. Samples containing more than one ionizable component were often unusable. The limitations of ion suppression led to the development of a new, more generally applicable approach to separation of ionized components: ion pair chromatography.

Developed by Dr. Gordon Schill in 1973, ion pair chromatography relies upon the addition of ionic compounds to the mobile phase to promote the formation of ion pairs with charged analytes. These reagents are comprised of an alkyl chain with an ionizable terminus (figure 1). When used with common hydrophobic HPLC phases in the reversed-phase mode, ion pair reagents can be used to selectively increase the retention of charged analytes (figure 2).

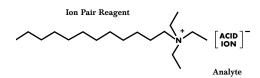


Figure 1. Quaternary Amine (Q-Series) Ion Pair Reagent.

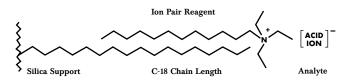
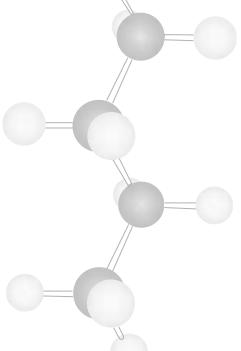


Figure 2. Quaternary Amine (Q-Series) Ion Pair Reagent interacting with C-18 Support.



ULTRAPURE ION PAIRING REAGENTS AND BUFFERS

Although ion exchange chromatography has become a popular mode of separation, it is not useful in all situations. The advantages of ion pair chromatography over ion exchange chromatography are:

- Simple preparation of buffers
- Wide choice of carbon chain lengths for improved retention and separation
- · Significantly reduced separation time
- Simultaneous separation of both ionized and nonionized solutes
- Highly reproducible results
- Improved peak shape

Regis Provides a Choice of Reagents

Regis manufactures both ultrapure anionic Sulfonate (S-Series) and cationic Quaternary Amine (Q-Series) ion pair concentrates in the following alkyl chain lengths: pentyl, hexyl, heptyl, octyl, and dodecyl. Alkyl chains are represented by cardinal numbers in the naming of our products, i.e., 5, 6, 7, 8, and 12. (See product descriptions on the following pages.)

Optical A	Optical Absorbance (AUFS)			bsorbance	(AUFS)
S-Series	200 nm	210 nm	Q-Series	200 nm	210 nm
S5	0.006	0.002	Q5	0.060	0.001
S6	0.048	0.018	Q6	0.059	0.006
S7	0.008	0.001	Q7	0.022	0.009
S8	0.001	0.003	Q8	0.082	0.003
S12	0.002	0.003	Q12	0.102	0.013
CH3CN	0.076	0.013			
CH3OH	0.940	0.510			

Table 1. Typical optical absorbances (AUFS) at 0.005 M.

Purity is a Key Ingredient

Purity is of key importance in the manufacture of our Ion Pair Reagents. Regis S- and Q-Series products are synthesized in accordance with the industry's highest quality standards, resulting in exceptional purity and integrity. This is demonstrated in table 1: UV transparency as low as 200 nm can be achieved for both the S- and Q-Series reagents. In most cases, these absorbances are lower than those for HPLC grade acetonitrile and methanol.

Although the S- and Q-Series ion pair reagents can be used at wavelengths less than 210 nm, the crucial factors in determining what wavelength to use are the integrity of the detector optics and the purity of the organic modifiers.

Regis also supplies bulk Sulfonate and several additional bulk Ion Pair Reagents and Rivier Buffers to complement the separation capabilities of the Sulfonate S-Series and Quaternary Amine Q-Series.

ULTRAPURE ION PAIRING REAGENTS AND BUFFERS

How to Select a Regis Ion Pair Reagent For Method Development

To choose the proper reagent, alkyl chain lengths must be taken into consideration. The chain lengths enable selective separation of the analyte. The longer the chain, the more hydrophobic the counterion, and therefore, greater the retention. Retention may increase by a factor of almost 20 when going from pentyl (Q5) to dodecyl (Q12), as illustrated in table 2 and figure 3. Both table 2 and figure 3 demonstrate that the Q-reagent chain length governs benzoic acid retention times, but does not affect the benzyl alcohol retention times. Similar behavior can also be achieved with the S-Series.

The following are guidelines to developing a successful method using Regis' ion pair reagents:

- Select a column endcapped ODS (octadecylsilyl) is most common.
- Use only HPLC-grade water and chromatography grade reagents in mobile phase preparation.
- Choose the mobile phase components and concentrations that give the best separation.
- If nonionic components are present in the sample, optimize the resolution prior to attempting ionic separations.
- Select the appropriate ion pair series to provide the necessary counterion. Use the Q-series for acidic compounds and the S-series for basic compounds.
- Through a process of elimination, choose the alkyl chain length which results in the best separation (figure 4).
- Once the reagent has been selected, adjust the pH of the mobile phase to maximize resolution. Because slight modification of pH can profoundly effect retention and selectivity, make all adjustments in small increments and monitor carefully (table 3).
- Ideally, the ion pair reagent concentration in the mobile phase should be 0.005 M. However, small adjustments in reagent concentration may increase retention slightly and optimize the separation (figure 5).

	Retention	Retention Ratio	
Q-Series	Benzoic Acid Benzyl Alcohol		Acid/Alcohol
Q5	4.53	9.17	0.49
Q6	6.50	8.60	0.76
Q7	8.24	9.13	0.90
Q8	12.36	8.94	1.38
Q12	<i>7</i> 9.53	8.52	9.33

Table 2. Retention vs. chain length.

[benzoic acid/benzyl alcohol in (60/40) water/methanol]						
G	16	Q	7	Q	8	
рΗ	R	рН	R	рН	R	
<i>7</i> .50	0.59	<i>7</i> .50	0.88	<i>7</i> .51	1.06	
6.50	0.70	6.51	1.00	6.54	1.29	
5.50	0.96	5.52	1.23	5.50	1.59	

Table 3. Retention ratio R as a function of pH.

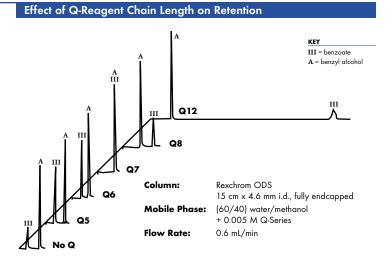


Figure 3. Retention increases with Q-Reagent chain length.

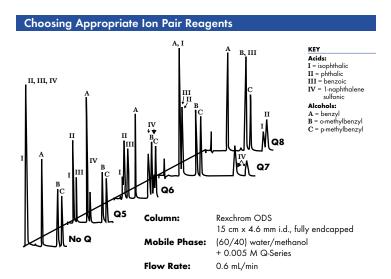


Figure 4. In a mixture of ionic and nonionic compounds, first separate the nonionic compounds from each other (See above). Then choose the ion pair reagent that retains the ionic compounds as desired. Here, Q6 seems to be the reagent of choice since all peaks are visibly separated.

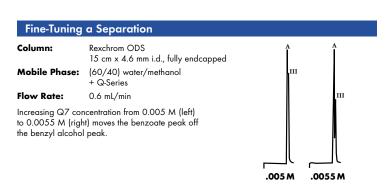


Figure 5. Increasing the Q-Reagent concentration may increase retention slightly and optimize the separation.

REGIS SULFONATES (S-SERIES) FOR BASIC COMPOUNDS

S-Series Ion Pair Concentrates (For Cations)

The sulfonates are sodium salts that act as an anionic counterion for the separation and resolution of positively charged analytes. The sulfonates are available as:

Ion pair concentrates — premixed 0.5 M solutions of alkyl sulfonates. When diluted to 1 L with HPLC-grade water, a 10 mL bottle forms a 0.005 M solution.

Larger quantities are available upon request. Please call Regis for pricing.

Product	Size	Product #	U.S. Price
S5 (1-pentylsodiumsulfonate)	(5) 10 mL bottles	405025	\$72.00
	100 mL bottle	405035	\$105.00
S6 (1-hexylsodiumsulfonate)	(5) 10 mL bottles	405026	\$72.00
	100 mL bottle	405036	\$105.00
S7 (1-heptylsodiumsulfonate)	(5) 10 mL bottles	405027	\$72.00
	100 mL bottle	405037	\$105.00
S8 (1-octylsodiumsulfonate)	(5) 10 mL bottles	405028	\$72.00
	100 mL bottle	405038	\$105.00
S12 (1-dodecylsodiumsulfonate)	(5) 10 mL bottles	405021	\$72.00
	100 mL bottle	405031	\$105.00

0.5 M solutions of Alkyl Sulfonates

(Each 10 mL bottle, diluted to 1 L, produces a 0.005 M solution)

S-Series Method Development Kit

Each kit contains a 10 mL bottle of each of the following:

CE C4 C7 C0 C10	405020	44000	
S5, S6, S7, S8, S12	405020	\$68.00	

Bulk Io	ı Pair	Reagents
(For Cat	ions)	

Bulk powder – fine, purified crystals, for use as a buffer in large-scale mobile phase preparation.

Larger quantities are available upon request. Please call Regis for pricing.

Product	Size	Product #	U.S. Price
1-Pentanesulfonate,			
Sodium Salt	25 gm	403025	\$55.00
	100 gm	403125	\$150.00
1-Hexanesulfonate,			
Sodium Salt	25 gm	403026	\$55.00
	100 gm	403126	\$150.00
1-Heptanesulfonate,			
Sodium Salt	25 gm	403027	\$55.00
	100 gm	403127	\$150.00
1-Octanesulfonate,			
Sodium Salt	25 gm	403028	\$55.00
	100 gm	403128	\$150.00
1-Dodecanesulfonate,			
Sodium Salt	5 gm	403021	\$20.00
	25 gm	403022	\$80.00

ION PAIR APPLICATIONS

Separation of Catecholamines using Ion Pair Reagents

Column: REXCHROM ODS, 5μm, 100Å

15 cm x 4.6 mm i.d.

Mobile Phase: (89/11) 0.005 M S8 ion pair concentrate/

acetonitrile, pH 2.5

 $\begin{tabular}{lll} Flow Rate: & 1.0 mL/min \\ Load: & 10 ~\mu L \\ \hline \end{tabular}$ $\begin{tabular}{lll} Detection: & UV 280 ~nm \\ \hline \end{tabular}$

Peak Identities: A. 3,4-Dihydroxyphenylacetic acid

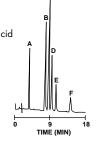
B. Norepinephrine

C. 3,4-Dihydroxyphenylalanine

D. Epinephrine

E. 3,4-Dihydroxybenzylamine

F. Dopamine



Separation of Artificial Sweeteners

Column: REXCHROM Little Champ II (ODS), 3µm, 100Å

5 cm x 4.6 mm i.d.

Mobile Phase: (86/14) 0.005 M Q7 ion pair concentrate/

acetonitrile, pH 6.2

 Flow Rate:
 1.0 mL/min

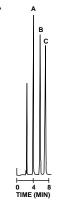
 Load:
 10 μL

 Detection:
 UV 210 nm

 Peak Identities:
 A. Aspartame

B. Acesulfame Potassium (Acesulfame-K)

C. Saccharin



REGIS QUATERNARY AMINES (Q-SERIES) FOR ACIDIC COMPOUNDS

Q-Series Ion Pair Concentrates (For Anions)

The Q-series is comprised of quaternary alkyltriethylamines that can be used for the resolution of negatively charged species. This unique set of cationic reagents was developed to complement the Sulfonate Series (S-Series) and is exclusively manufactured by Regis. The Quaternary Alkyltriethylamines are available as:

Ion pair concentrates — premixed 0.5 M solutions of alkyl amines. When diluted to 1 L with HPLC-grade water, a 10 mL bottle forms a 0.005 M buffered solution.

Product	Size	Product #	U.S. Price
Q5 (1-pentyltriethyl-			
ammonium phosphate)	(5) 10 mL bottles	404025	\$72.00
	100 mL bottle	404035	\$105.00
Q6 (1-hexyltriethyl-			
ammonium phosphate)	(5) 10 mL bottles	404026	\$72.00
	100 mL bottle	404036	\$105.00
Q7 (1-heptyltriethyl-			
ammonium phosphate)	(5) 10 mL bottles	404027	\$72.00
	100 mL bottle	404037	\$105.00
Q8 (1-octyltriethyl-			
ammonium phosphate)	(5) 10 mL bottles	404028	\$72.00
	100 mL bottle	404038	\$105.00
Q12 (1-dodecyltriethyl-			
ammonium phosphate)	(5) 10 mL bottles	404021	\$72.00
	100 mL bottle	404031	\$105.00

0.5 M solutions of Quaternary Alkyltriethylamines (Each 10 mL bottle, diluted to 1 L, produces a 0.005 M solution)

Q-Series Method Development Kit

Each kit contains a 10 mL bottle of each of the following:

Q5, Q6, Q7, Q8, Q12 404020 \$68.00

Other Regis Bulk Ion Pair Reagents (For Anions)

Other bulk Ion Pair reagents such as Tetrabutylammonium phosphate, Trihexylamine and Triheptylamine are complementary reagents used for the resolution of negatively charged analytes.

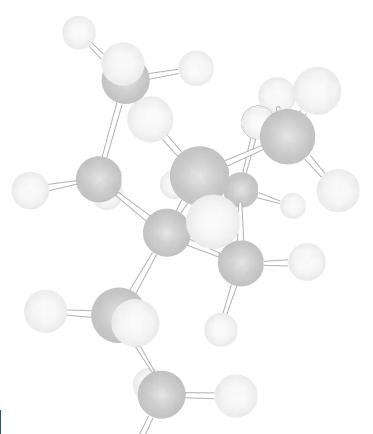
Product	Size	Product #	U.S. Price
Tetrabutylammonium phosphate 0.5 M, pH 7.5	10 mL	680502	\$15.00
Tetrabutylammonium phosphate 0.5 M, pH 7.5	500 mL	680503	\$270.00

Ion Pair References

- Perry, J. A.; Glunz, L. G.; Szczerba, T. J.; Hocson, V. S.; Reagents For Ion Pair Reversed-Phase HPLC; American Laboratory 1984, 16(10), 114–119.
- Eksborg, S.; Lagerstrom, P.; Modin, R.; Schill, G.; Ion Pair Chromatography of Organic Compounds J. Chrom. 1973, 83, 99.
- Eksborg, S.; Schill, G.; Ion Pair Partition Chromatography of Organic Ammonium Compounds Anal. Chem. 1973, 45, 2092.

For additional information on Ion Pair Reagents, check our Web site at www.registech.com/ionpair/ or contact Regis directly at:

(800) 323-8144 ext. 649 (847) 967-6000 ext. 649 e-mail us at: sales@registech.com.





DERIVATIZATION

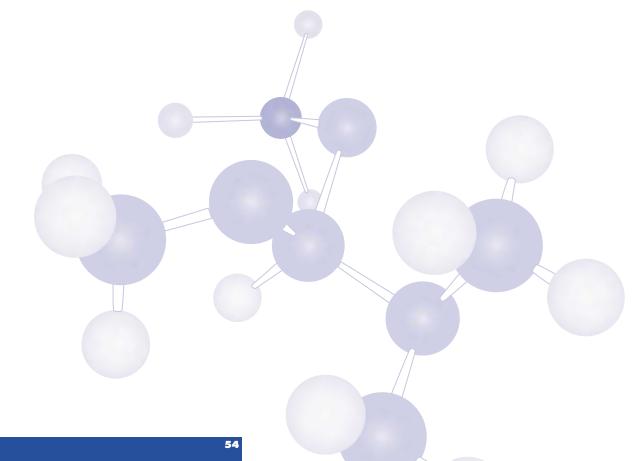
Derivatization

Derivatization is primarily performed to modify an analyte's functionality in order to enable chromatographic separations. For more than 30 years, Regis has been a leader in the manufacture of highly pure derivatization reagents for gas chromatography.

The formation of chemical derivatives to facilitate meaningful analysis has long been a common practice in gas chromatography. For the analytical chemist, judicious use of derivatization can be the key to unlocking and simplifying a great many complex and perplexing separation problems. Derivatization, accomplished through alteration of functional groups, provides:

- · Increased sample volatility
- Improved selectivity and chromatographic efficiency
- Enhanced detectability

Additional procedures and references are available on our Web site at: www.registech.com.



Sample volatility or thermal stability is crucial in GC applications. If a sample does not possess these important characteristics, GC analysis is highly unproductive. Derivatization techniques have been developed to address these issues to insure successful separations. In GC Derivatization, replacement of active hydrogen in functional groups, such as -COOH, -OH, -NH, and -SH, is the primary area of concern and is accomplished through either silylation, acylation or alkylation.

GAS CHROMOTOGRAPHY (GC) DERIVATIZATION

Silylation

Silylation is the most widely used derivatization procedure for sample analysis by GC. Silylation reagents are popular because they are easy to use and readily form derivatives. In silylation, an active hydrogen is replaced by an alkylsilyl group, such as trimethylsilyl (TMS) or t-butyldimethylsilyl (t-BDMS). Compared to their parent compounds, silyl derivatives are more volatile, less polar, and more thermally stable. As a result, GC separation is improved and detection is enhanced.

Silylation reagents are generally moisture sensitive, requiring them to be sealed under nitrogen to prevent deactivation. The derivatives of TMS reagents are also moisture sensitive. In response to this difficulty, t-BDMS reagents were introduced, which enabled the formation of derivatives 10,000 times more stable to hydrolysis than the TMS ethers.

Both TMS and t-BDMS reagents are suitable for a wide variety of compounds, offer excellent thermal stability and can be used in a variety of GC conditions and applications.

Analysis by the popular combination of gas chromatography and mass spectrometry (GS/MS) often requires special sample derivatization. Particularly effective in these applications are MTBSTFA and deutero-MTBSTFA-d.



Acylation

Acylation reagents offer the same types of advantages available from silylation reagents: creating less polar, more volatile derivatives. However, in comparison to silylating reagents, the acylating reagents more readily target highly polar, multi-functional compounds, such as carbohydrates and amino acids. In addition, acylating reagents provide the distinct advantage of introducing electron-capturing groups, thus enhancing detectability during analysis.

Generally, these reagents are available as acid anhydrides, acyl derivatives, or acyl halides. The acyl halides and acyl derivatives are highly reactive and are suitable for use where steric hindrance may be a factor. Acid anhydrides are supplied in a number of fluorinated configurations, which improve detection. These fluorinated anhydride derivatives are used primarily for Electron Capture Detection (ECD), but can also be used for Flame Ionization Detection (FID). Fluorinated anhydrides are often used in derivatizing samples to confirm drugs of abuse. Despite the special utility of these reagents, their acidic nature requires that any excess or byproducts be removed prior to

analysis to prevent deterioration of the column.



Alkylation

As with other derivatization reagents, alkylation reagents reduce molecular polarity by replacing active hydrogens with an alkyl group. These reagents are used to modify compounds having acidic hydrogens, such as carboxylic acids and phenols. Alkylation reagents can be used alone to form esters, ethers, and amides—or they can be used in conjunction with acylation or silylation reagents. A two-step approach is commonly used in the derivatization of amino acids,

where multiple functional groups on these compounds may necessitate protection during derivatization.

Due to the availability of reagents and their ease of use, esterification (the reaction of an acid with an alcohol in the presence of a catalyst to form an ester) is the most popular method of alkylation. Alkylation reagents are available in several configurations that enable the formation of a variety of esters. Alkyl esters are stable, and can be formed quickly and quantitatively. By altering the length of the substituted alkyl group, retention of the derivative can be varied. In addition to the formation of simple esters, alkylation reagents

can be used in extractive procedures where biological matrices may be present.

GC Chiral Derivatization

GC analysis of enantiomeric compounds on nonracemic or achiral stationary phases requires the use of enantiopure derivatization reagents. These reagents generally target one specific functional group to produce diastereomers of each of the enantiomeric analytes. From the resulting chromatograms, calculations are conducted to determine the enantiomeric concentration of the analyte.



GUIDE TO GC DERIVATIZATION METHODS/FUNCTIONAL GROUPS

	GC Derivatization Method			
Functional Group	Silylation	Acylation	Alkylation	
Active Hydrogens	BSA, BSTFA, BSTFA/TMCS, Deriva-Sil, Hydrox-Sil, MSTFA, MTBSTFA, TMSI	PFPOH/PFPA	DMF Dialkylacetals, TBH	
Carboxylic Acids	BSTFA, Hydrox-Sil Conc., MTBSTFA, TMSI	PFPOH/PFPA	BF ₃ /Methanol, BF ₃ /n-Butanol, DMF Dialkylacetals, TBH	
Alcohols and Phenols: unhindered and moderately hindered	BSA, BSTFA/TMCS, HMDS, MTBSTFA/t-BDMCS	HFBI, Fluorinated anhydrides (HFBA, PFPA, TFAA), MBTFA, MCF*	DMF Dialkylacetals, PFB-Br/TBA-H-SO₄, TBH	
Alcohols and Phenols: highly hindered	BSTFA/TMCS, Deriva-Sil, Deriva-Sil Conc.	Fluorinated anhydrides (HFBA, PFPA, TFAA), HFBI, PFBCI	DMF Dialkylacetals, PFB-Br/TBA-H-SO₄, TBH	
Amines: primary and secondary	BSTFA, MTBSTFA/t-BDMCS	Fluorinated anhydrides (HFBA, PFPA, TFAA), HFBI, MBTFA, PFBCI, TPC*	DMF Dialkylacetals, TBH	
Amides	BSA, BSTFA, BSTFA/TMCS, Deriva-Sil Conc.	HFBI	DMF Dialkylacetals, TBH	
Amino Acids	BSTFA, TMSI	HFBI (+ Silylation)	DMF Dialkylacetals, TBH	
Catecholamines	TMSI	Fluorinated anhydrides (HFBA, PFPA, TFAA), HFBI		
Carbohydrates and Sugars	HMDS, Hydrox-Sil AQ, TMSI	мвтға		
Inorganic Anions	BSTFA, MTBSTFA			
Nitrosamines		HFBA		
Sulfonamides	BSTFA	Fluorinated anhydrides (HFBA, PFPA, TFAA)	DMF Dialkylacetals, PFB-Br/TBA-H-SO ₄	
	Derivatization reagents are listed in alphabetical order, not in order of preference.	*For Chiral Analysis	Source: Knapp, D.R. Handbook of Analytical Derivatization Reactions; John Wiley and Sons: New York, 1979.	

SILYLATION REAGENTS

BSA

N,O-Bis(trimethylsilyl)acetamide

• Forms highly stable TMS derivatives with most organic functional groups under mild reaction conditions.

Product	Size	Product #	U.S. Price
BSA	10 x 1 gm	270501	\$48.00
	4 x 5 gm	270502	\$48.00
	25 gm	270503	\$56.00
	100 gm	270504	\$134.00

O—TMS

$$\downarrow$$

 H_3C — C = N —TMS + H — Y — R \longrightarrow TMS— Y — R + H_3C — C — N —TMS
 \downarrow
TMS = Si(CH₃)₃ Y = O, S, NH, NR¹, COO

 $R, R^1 = Alk, Ar$

BSTFA-Regisil® BSTFA +TMCS (1%, 10%) N,O-Bis(trimethylsilyl)trifluoroacetamide

- Reacts faster and more completely than BSA, due to presence of trifluoroacetyl group.
- The high volatility of BSTFA and its byproducts results in separation of early eluting peaks.
- Highly volatile and stable products result in low detector noise and fouling.
- Excellent solubility.
- Addition of TMCS catalyzes reactions of hindered functional groups in secondary alcohols and amines.

Product	Size	Product #	U.S. Price
Regisil® RC-1	10 x 1 gm	270111	\$66.00
(BSTFA)	50 x 1 gm	27011 <i>7</i>	\$225.00
	4 x 5 gm	270112	\$59.00
	25 gm	270113	\$66.00
	100 gm	270114	\$180.00
Regisil® RC-2	10 x 1 gm	270121	\$66.00
(BSTFA) +1% TMCS	50 x 1 gm	270127	\$225.00
	4 x 5 gm	270122	\$59.00
	25 gm	270123	\$66.00
	100 gm	270124	\$180.00
Regisil® RC-3	10 x 1 gm	270131	\$66.00
(BSTFA) +10% TMCS	4 x 5 gm	270132	\$59.00
	25 gm	270133	\$66.00
	100 gm	270134	\$180.00

HMDS

Hexamethyldisilazane

- Weak TMS donor, used for silylation of carbohydrates.
- Used as mixture with pyridine and trifluoroacetic acid.

Product	Size	Product #	U.S. Price
HMDS	25 gm	270651	\$27.00
	100 gm	270652	\$49.00

N-Methyltrimethylsilyltrifluoroacetamide

- Most volatile of the TMS-acetamides.
- Useful in the analysis of volatile trace materials.

Product	Size	Product #	U.S. Price
MSTFA	10 x 1 gm	270590	\$65.00
	10 gm	270589	\$60.00
	25 gm	270593	\$ <i>7</i> 6.00
	100 gm	270594	\$265.00

Regisil is a registered trademark of Regis Technologies, Inc.

SILYLATION REAGENTS

MTBSTFA

MTBSTFA + 1% t-BDMCS

N-Methyl-N-(t-butyldimethylsilyl)trifluoroacetamide

- Replaces active hydrogens to form t-BDMS derivatives.
- Derivatization is usually complete upon dissolution with this exceptionally strong, yet mild silylating reagent.
- MTBSTFA derivatives are 10⁴ times more stable to hydrolysis than their corresponding TMS derivatives.
- Produces easily interpreted mass spectra for GC/MS.
- Addition of t-BDMCS catalyzes reactions of hindered alcohols and amines.

Product	Size	Product #	U.S. Price
MTBSTFA (+1% t-BDMCS)	5 x 1 gm 10 x 1 gm 2 x 5 gm 25 gm	270141 270144 270142 270143	\$81.00 \$125.00 \$83.00 \$150.00
MTBSTFA (no t-BDMCS)	5 x 1 gm 2 x 5 gm 25 gm	270241 270242 270243	\$81.00 \$83.00 \$150.00

TMCS

Trimethylchlorosilane

 Used as a catalyst to increase reactivity of other silylation reagents.

Product	Size	Product #	U.S. Price
TMCS	25 gm	270601	\$27.00
	100 gm	270602	\$49.00

$$CH_3$$
 CH_3 CH_3

TMSI

Trimethylsilylimidazole

- Potent, selective TMS donor that reacts with alcohols and phenols but not amines or amides.
- Derivatizes wet sugar samples, hindered hydroxyl groups in steroids, and amino acids in fluorinated acylation reagents.
- Used in the preparation of dual perfluoroacyl and TMS derivatives.

Product	Size	Product #	U.S. Price
TMSI	10 x 1 gm	270401	\$47.00
	5 gm	270402	\$25.00
	25 gm	270403	\$62.00

SILYLATION FORMULATIONS

Deriva-Sil

- BSTFA:TMCS:TMSI:Pyridine (3:2:3:10) formulation.
- Derivatizes sterically-hindered compounds.
- Reacts with carbohydrates, hydroxy- and keto-steroids, fatty acids, and some amines and amides.
- Derivatizations are complete in minutes.

Product	Size	Product #	U.S. Price
Deriva-Sil — BSTFA:TMCS: TMSI:Pyridine (3:2:3:10)	10 x 1 mL 25 mL	270151 270152	\$65.00 \$65.00

Deriva-Sil Concentrate

- BSTFA:TMCS:TMSI (3:2:3) concentrate formulation.
- Used for applications where pyridine is undesirable (i.e., 3-ketosteroids).

Product	Size	Product #	U.S. Price
Deriva-Sil Concentrate – BSTFA:TMCS:TMSI (3:2:3)	25 mL	270150	\$86.00

SILYLATION FORMULATIONS

Hydrox-Sil

- HMDS:TMCS:Pyridine (2:1:10) formulation for one-step derivatizations.
- Fast formation of the TMS derivatives of organic acids, unhindered alcohols and phenols, and some amines.

Product	Size	Product #	U.S. Price
Hydrox-Sil — Reagent HMDS:TMCS:Pyridine (2:1:10)	10 x 1 mL 25 mL	270455 270457	\$52.00 \$52.00

Hydrox-Sil AQ

- TMSI:Pyridine (21%w/v) formulation.
- Forms TMS derivatives of hydroxyl and polyhydroxyl compounds in the presence of water.

Product	Size	Product #	U.S. Price
Hydrox-Sil AQ –	10 x 1 mL	270451	\$54.00
TMSI:Pyridine	25 mL	270453	\$50.00
(21% w/v)			

Hydrox-Sil Concentrate

- HMDS:TMCS (2:1) concentrate formulation.
- Suited for applications where pyridine in Hydrox-Sil is undesirable.

Product Size	Product #	U.S. Price
Hydrox-Sil Concentrate – 25 mL	270458	\$76.00
HMDS:TMCS (2:1)		

ACYLATION REAGENTS

Fluorinated Anhydrides: HFBA/PFPA/TFAA

Heptafluorobutyric Anhydride/Pentafluoropropionic Anhydride/Trifluoroacetic Anhydride

- Most commonly used for ECD.
- Reacts with alcohols, amines, and phenols.
- Bases such as triethylamine and trimethylamine can be added to promote reactivity.
- Frequently used for the confirmation of drugs of abuse.
- HFBA derivatives are the most sensitive to ECD.
- PFPA derivatives require the lowest analysis temperatures.
- TFAA is the most reactive and volatile of the anhydrides.

Size	Product #	U.S. Price	
10 x 1 gm	270851	\$68.00	
25 gm	270853	\$91.00	
10 x 1 gm	640110	\$85.00	
25 gm	640113	\$90.00	
100 gm	640114	\$250.00	
10 x 1 gm	270841	\$56.00	
25 gm	270843	\$26.00	
	10 x 1 gm 25 gm 10 x 1 gm 25 gm 100 gm 10 x 1 gm	10 x 1 gm 270851 25 gm 270853 10 x 1 gm 640110 25 gm 640113 100 gm 640114 10 x 1 gm 270841	10 x 1 gm 270851 \$68.00 25 gm 270853 \$91.00 10 x 1 gm 640110 \$85.00 25 gm 640113 \$90.00 100 gm 640114 \$250.00 10 x 1 gm 270841 \$56.00

TFAA
$$F_3C$$
- C - O - C - CF_3 + H-Y-R \longrightarrow F_3C - C -Y-R + F_3C - C - O -P

Y = O, NH, NR¹R, R¹ = Alk, Ar

HFBI

Heptafluorobutyrylimidazole

- Readily forms derivatives with phenols, alcohols and amines suitable for ECD.
- Reactions are fast and mild.
- Imidazole is not acidic, so no decomposition or corrosion occurs on columns.

ACYLATION REAGENTS

MBTFA

N-Methyl-N-bis(trifluoroacetamide)

- Reacts rapidly under mild conditions with primary and secondary amines.
- Reacts more slowly with alcohols, phenols, and thiols.
- Works well in the analysis of sugars.

Product	Size	Product #	U.S. Price
MBTFA	10 x 1 gm	270092	\$84.00
	5 gm	270091	\$39.00
	25 gm	270095	\$160.00
	100 gm	270093	\$395.00

PFPOH

2,2,3,3,3-Pentafluoropropanol

- Used in combination with PFPA to make derivatives of the most common functional groups, especially polyfunctional bio-organic compounds.
- Formed derivatives are highly suitable for ECD.

Product	Size	Product #	U.S. Price
PFPOH	5 gm	270815	\$32.00
	25 gm	270816	\$100.00

DERIVATIZATION GRADE SOLVENTS

Derivatization Grade Solvents

- High purity reagents packaged under nitrogen.
- Sealed with Teflon®-coated septa, allowing easy access to sample without exposure to moisture and oxygen.

Product	Size	Product #	U.S. Price
Acetonitrile	2 x 25 mL	270010	\$20.00
Dimethylformamide	2 x 25 mL	270011	\$20.00
Pyridine	2 x 25 mL	270013	\$20.00
Tetrahydrofuran	2 x 25 mL	270014	\$20.00

GC CHIRAL AND SPECIALTY DERIVATIZATION REAGENTS

TPC

N-Trifluoroacetyl-L-Prolyl Chloride

- Couples with amines to form diastereomers which can be separated on GC columns.
- Provides sample volatility.
- Used for confirmation of drugs of abuse testing.

Product	Size	Product #	U.S. Price
TPC	25 mL	440001	\$94.00

MCF

(1R, 2S, 5R)-(-)-Menthylchloroformate

• Resolves enantioenriched alcohols.

Product	Size	Product #	U.S. Price	
MCF	25 mL	440003	\$95.00	

$$CH_3$$
 CH_3
 CH_3

Teflon is a registered trademark of DuPont Company.

GC CHIRAL AND SPECIALTY DERIVATIZATION REAGENTS

HFIP

1,1,1,3,3,3-Hexafluoro-2-Propanol

 Esterification reagent for the determination of aromatic acids in tissue by GC and electron capture detection.

Product	Size	Product #	U.S. Price
HFIP	10 gm	270701	\$22.00
	25 gm	270702	\$40.00
	4 x 25 gm	270703	\$120.00
	100 gm	270704	\$120.00

3.0 N HCL in n-Butanol

 Most commonly used for rapid diagnosis of neonatal blood spots by Tandem Mass Spectrometry.

Product	Size	Product #	U.S. Price
3.0 N HCL in n-Butanol	4 x 25 mL	201007	\$44.00
	100 mL	201009	\$44.00
	500 mL	201010	\$225.00

Col-Treet

- Silylating mixture designed for conditioning gas chromatography columns.
- Eliminates peak tailing, peak broadening and residues which may reduce column efficiency.
- Not to be used with any hydrogen bonding stationary phases such as CARBOWAX® or phases with active hydrogen sites.

Product	Size	Product #	U.S. Price
Col-Treet	10 x 1 mL	970130	\$40.00
	4 x 5 mL	970131	\$50.00

Glas-Treet

- Used to deactivate active sites found in glassware, chromatography columns, and glass wool.
- Does not coat glass; renders it inactive chemically.

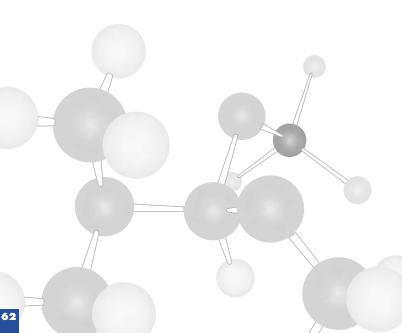
 Product
 Size
 Product #
 U.S. Price

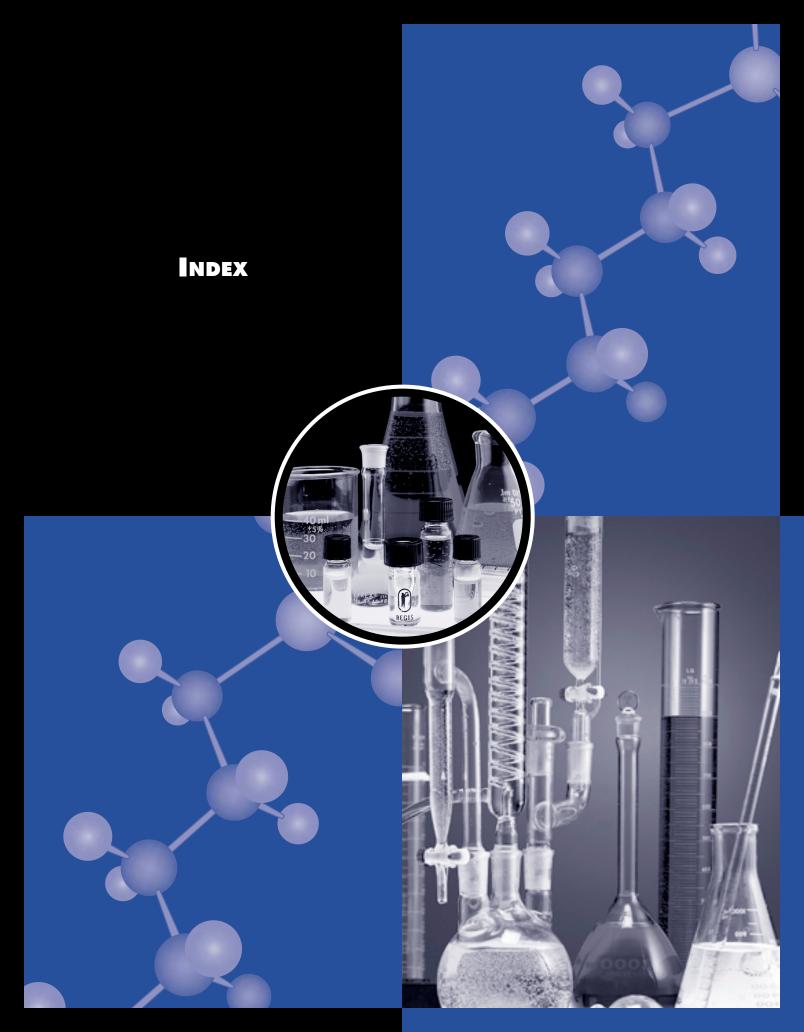
 Glas-Treet
 100 mL
 970068
 \$18.00

CARBOWAX is a registered trademark of Union Carbide Corp.

For additional information on GC Derivatization Reagents, check our Web site at www.registech.com/gc/or contact Regis directly at:

(800) 323-8144 ext. 649 (847) 967-6000 ext. 649 e-mail us at: sales@registech.com.





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Morton Grove, IL 60053-0519 USA

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Shipping address

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Purchase order number

Product #

Quantity

Contact person

Phone number

Fax number

Payment Details

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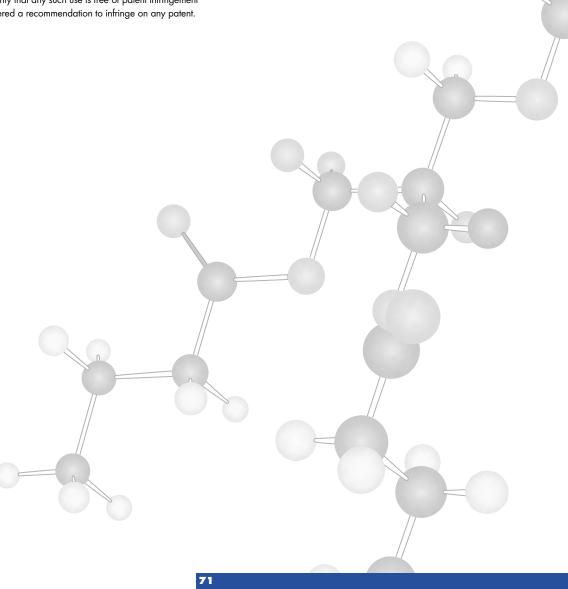
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