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# TECHNICAL NOTE 2. Preparation of mobile phase for HPLC

## 1) Organic solvent/ aqueous mixed mobile phase

## 1)-1. Preparation of methanol : water = 70 : 30 (v/v) 1L

1 Measure 700 ml of methanol in a measuring cylinder.

2 Measure 300 ml of distilled water in a measuring cylinder.

3 Mix 1 and 2 thoroughly and degas.

The better approach is to prepare the mobile phase gravimetrically rather than volumetrically. Following is example of preparation.

Composition table for	mobile phase	1L using methanol	and water
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Methanol / Water	Methanol (g)	Distilled water (g)
90/10 (v/v)	711.9	99.8
80/20 (v/v)	632.8	199.6
70/30 (v/v)	553.7	299.5
60/40 (v/v)	474.6	399.3
50/50 (v/v)	395.5	499.1
40/60 (v/v)	316.4	598.9
30/70 (v/v)	237.3	698.7
20/80 (v/v)	158.2	798.6
10/90 (v/v)	79.1	898.4

Composition table for mobile phase 1L using acetonitrile and water

Acetonitrile / Water	Acetonitrile (g)	Distilled water (g)	
90/10 (v/v)	707.4	99.8	
80/20 (v/v)	628.8	199.6	
70/30 (v/v)	550.2	299.5	
60/40 (v/v)	471.6	399.3	
50/50 (v/v)	393.0	499.1	
40/60 (v/v)	314.4	598.9	
30/70 (v/v)	235.8	698.7	
20/80 (v/v)	157.2	798.6	
10/90 (v/v)	78.6	898.4	

Caution : Methanol and acetonitrile are hazardous substances, do not use for medical purpose. Always process in a laboratory hood and wear an eye protection and a mask.





 Column
 COSMOSIL 5C18-MS-II

 Column size
 4.6 mm I.D. × 150 mm

 Flow rate
 1.0 ml/min

 Detection
 254 nm 0.16 AUFS

 Temperature
 30°C

 Sample
 1 uracil

 2. naphthalene

Special attention should be paid to measure correct amount of organic solvent as the retention time is significantly changed by 1% different composition.

Differences between acetonitrile and methanol in reversed phase liquid chromatography





## 2) Organic solvent/ buffer mixed mobile phase

- 2)-1. Preparation of methanol : 20 mmol/l phosphate buffer (pH2.5) = 80 : 20 (v/v) 1L
  - Preparation of 20 mmol/l phosphate buffer (pH2.5)
  - 1 Prepare 20 mmol/l sodium dihydrogenphosphate aqueous solution.
  - ② Prepare 20 mmol/l phosphoric acid aqueous solution.
  - 3 Adjust the pH to 2.5 by mixing 1 with 2.
  - ④ Filter ③ under reduced pressure to remove insoluble substance that may deteriorate pump-seal and clog columns (0.45 μm or smaller pore size is recommended).

### (Easy method)

- ① Dissolve 1.31 g of sodium dihydrogenphosphate and 1.05 g of phosphoric acid in distilled water to make 1 L solution.
- (2) Filter the solution under reduced pressure to remove insoluble substance (0.45  $\mu$ m or smaller pore size is recommended).
- 3 Confirm that the solution is pH2.5.
- Preparation of methanol : 20 mmol/l phosphate buffer (pH2.5) = 80 : 20 1L
- ① Measure 800 ml of methanol in a measuring cylinder.
- 0 Measure 200 ml of 20 mmol/l phosphate buffer (pH2.5) in a measuring cylinder.
- 3 Mix 1 and 2 thoroughly and degas.

### 2)-2. Preparation of methanol : 20 mmol/l phosphate buffer (pH7.0) = 80 : 20 (v/v) 1L

- Preparation of 20 mmol/I phosphate buffer (pH7.0)
- ① Prepare 20 mmol/l sodium dihydrogenphosphate aqueous solution.
- (2) Prepare 20 mmol/l di-sodium hydrogenphosphate aqueous solution.
- 3 Adjust the pH7.0 by mixing 1 with 2 .
- ④ Filter ③ under reduced pressure to remove insoluble substance that may deteriorate pump-seal and clog columns (0.45 μm or smaller pore size is recommended).

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(Easy method)

- ① Dissolve 1.14 g of sodium dihydrogenphosphate and 1.49 g of di-sodium hydrogenphosphate in distilled water to make 1L solution.
- 2 Filter the solution under reduced pressure to remove insoluble substance (0.45  $\mu$ m or smaller pore size is recommended).
- 3 Confirm that the solution is pH7.0.
- Preparation of Methanol : 20 mmol/l phosphate buffer (pH7.0) = 80 : 20 1 L
- 1 Measure 800 ml of methanol in a measuring cylinder.
- 2 Measure 200 ml of 20 mmol/l phosphate buffer (pH7.0) in a measuring cylinder.
- 3 Mix 1 and 2 thoroughly and degas.

The better approach is to prepare the mobile phase gravimetrically rather than volumetrically. Following is example of preparation.

methanol : 20 mmol/l phosphate buffer	methanol (g)	20 mmol/l phosphate buffer (pH2.5) (g)	20 mmol/l phosphate buffer (pH7.0) (g)
90/10 (v/v)	711.9	99.8	99.9
80/20 (v/v)	632.8	199.6	199.8
70/30 (v/v)	553.7	299.4	299.7
60/40 (v/v)	474.6	399.2	399.6
50/50 (v/v)	395.5	499.0	499.5
40/60 (v/v)	316.4	598.8	599.4
30/70 (v/v)	237.3	698.6	699.3
20/80 (v/v)	158.2	798.4	799.2
10/90 (v/v)	79.1	898.2	899.1

Caution : Methanol and acetonitrile are hazardous substances, do not use for medical purpose. Always process in a laboratory hood and wear an eye protection and a mask.

### 3) Preparation of ion pair reagent containing mobile phase

3)-1.Preparation of 5 mmol/l sodium 1-butanesulfonate containing 20 mmol/l phosphate buffer (pH2.5)

- ① Prepare 5 mmol/l sodium 1-butanesulfonate containing 20 mmol/l sodium dihydrogenphosphate aqueous solution.
- ② Prepare 5 mmol/I sodium 1-butanesulfonate containing 20 mmol/I phosphoric acid aqueous solution
- 3 Adjust the pH to 2.5 by mixing 1 with 2 .
- ④ Filter ③ under reduced pressure to remove insoluble substance that may deteriorate pump-seal and clog columns (0.45 μm or smaller pore size is recommended).

(Easy method)

- ① Dissolve 1.31 g of sodium dihydrogenphosphate, 1.05 g of phosphoric acid and 0.80 g of sodium 1-butanesulfonate in distilled water to make 1L solution.
- (2) Filter the solution under reduced pressure to remove insoluble substance (0.45  $\mu$ m or smaller pore size is recommended).
- 3 Confirm that the solution is pH2.5.

% 0.5M sodium 1-butanesulfonate aqueous solution is also available from Nacalai Tesque.